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11 and THE REGENTS OF THE UNIVERSITY OF CALIFORNIA

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16 Attorneys for Plaintiffs and Counter-defendants LIFE  
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17 INC.

18 UNITED STATES DISTRICT COURT  
SOUTHERN DISTRICT OF CALIFORNIA

19 LIFE TECHNOLOGIES CORPORATION ,  
MOLECULAR PROBES, INC. AND THE  
20 REGENTS OF THE UNIVERSITY OF  
CALIFORNIA,

21 Plaintiffs-Counterclaim Defendants,

22 v.

23 EBIOSCIENCE INC,

24 Defendant-Counterclaim Plaintiff  
25

CASE NO. 10-CV-2127-IEG (NLS)

**PLAINTIFFS' FIRST AMENDED  
COMPLAINT AND JURY  
DEMAND**

1 Plaintiffs LIFE TECHNOLOGIES CORPORATION (“Life Technologies”),  
2 MOLECULAR PROBES, INC. (“Molecular Probes”), and THE REGENTS OF THE  
3 UNIVERSITY OF CALIFORNIA (“UC”), (herein collectively referred to as “Plaintiffs”)  
4 complain against Defendant EBIOSCIENCE INC. (“Defendant”) as follows:

5 **JURISDICTION AND VENUE**

6 1. This civil action for patent infringement arises under the patent laws of the United  
7 States, specifically under Title 35 of the United States Code, Sections 271, *et seq.* Subject matter  
8 jurisdiction in this Court is founded upon 28 U.S.C. §§ 1331 and 1338(a). The Court has personal  
9 jurisdiction over Defendant in that it is a California corporation with a principal place of business  
10 in this district. In addition, Defendant regularly conducts business in this district. In addition,  
11 Defendant regularly conducts business in this district and has committed acts in this judicial  
12 district that give rise to this action.

13 2. Defendant has committed acts of infringement within this judicial district giving  
14 rise to this action. Accordingly, venue is proper in this judicial district pursuant to 28 U.S.C. §§  
15 1391(b), (c) and/or 28 U.S.C. § 1400(b).

16 **REGARDING THE PARTIES**

17 3. Plaintiff Life Technologies is a Delaware corporation, with its principal place of  
18 business at 5791 Van Allen Way, Carlsbad, California 92009.

19 4. Plaintiff Molecular Probes is an Oregon corporation, with offices in Eugene,  
20 Oregon. Plaintiff Molecular Probes is a wholly owned subsidiary of Plaintiff Life Technologies.

21 5. Plaintiff UC is a public entity existing under the laws of the state of California.

22 6. Plaintiffs are informed and believe, and on that basis allege, that Defendant  
23 eBioscience Inc. is a California corporation that has its principal place of business in San Diego,  
24 California.

**FIRST CLAIM FOR RELIEF**  
**(Patent Infringement)**

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3 7. Plaintiffs repeat and reallege the allegations set forth in preceding paragraphs 1  
4 through 6, inclusive.

5 8. United States Letters Patent No. 6,423,551 (the “’551 Patent”), was duly and  
6 legally issued by the United States Patent and Trademark Office on July 23, 2002, and  
7 subsequently assigned to UC. Plaintiffs Life Technologies and Molecular Probes jointly hold a  
8 lawfully acquired, exclusive license to the ’551 Patent from UC. A true and correct copy of the  
9 ’551 Patent is attached hereto and incorporated herein by reference as Exhibit 1.

10 9. United States Letters Patent No. 6,699,723 (the “’723 Patent”), was duly and  
11 legally issued by the United States Patent and Trademark Office on March 2, 2004, and  
12 subsequently assigned to UC. Plaintiffs Life Technologies and Molecular Probes jointly hold a  
13 lawfully acquired, exclusive license to the ’723 Patent from UC. A true and correct copy of the  
14 ’723 Patent is attached hereto and incorporated herein by reference as Exhibit 2.

15 10. United States Letters Patent No. 6,927,069 (the “’069 Patent”), was duly and  
16 legally issued by the United States Patent and Trademark Office on August 9, 2005, and  
17 subsequently assigned to UC. Plaintiffs Life Technologies and Molecular Probes jointly hold a  
18 lawfully acquired, exclusive license to the ’069 Patent from UC. A true and correct copy of the  
19 ’069 Patent is attached hereto and incorporated herein by reference as Exhibit 3.

20 11. Defendant has been infringing, contributing to the infringement of, and/or inducing  
21 others to infringe the ’551 patent by making, manufacturing, promoting, marketing, advertising,  
22 distributing, offering for sale and selling and/or causing to be offered or sold certain eFluor®  
23 products that infringe one or more claims of the ’551 Patent literally and/or under the doctrine of  
24 equivalents.

25 12. Defendant has been infringing, contributing to the infringement of, and/or inducing  
26 others to infringe the ’723 patent by making, manufacturing, promoting, marketing, advertising,  
27 distributing, offering for sale and selling and/or causing to be offered or sold certain eFluor®

1 products that infringe one or more claims of the '723 Patent literally and/or under the doctrine of  
2 equivalents.

3 13. Defendant has been infringing, contributing to the infringement of, and/or inducing  
4 others to infringe the '069 patent by making, manufacturing, promoting, marketing, advertising,  
5 distributing, offering for sale and selling and/or causing to be offered or sold certain eFluor®  
6 products that infringe one or more claims of the '069 Patent literally and/or under the doctrine of  
7 equivalents.

8 14. Defendant's infringing products include, but are in no way limited to, Defendant's  
9 eFluor® Nanocrystals Products.

10 15. Defendant has continued its acts of infringement despite an objectively high  
11 likelihood of infringing valid and enforceable patents, namely the currently asserted '551, '723,  
12 and '069 Patents. Additionally, Defendant knew of the objectively high likelihood of infringing  
13 valid and enforceable patents and/or that risk was so obvious that it should have been known to  
14 Defendant. For example, in previous litigation in this district, Defendant alleged that it "has been  
15 and is making and using products pursuant to" an agreement between Defendant and Evident  
16 Technologies involving quantum dots. *eBioscience Corp. v. Invitrogen Corp.*, 08-cv-1729 JAH  
17 (LSP) (S.D. Cal.), DN 1, Compl. ¶ 13. Plaintiffs are informed and believe, and on that basis  
18 allege, that Defendant currently makes and uses the accused products pursuant to an agreement  
19 between Defendant and Evident Technologies.

20 16. At the time of the earlier litigation in this district, Evident Technologies was a  
21 defendant in litigation involving the '551, '723, and '069 Patents. *Invitrogen Corp. v. Evident*  
22 *Tech., Inc.*, 08-cv-163-LED/JDL (E.D. Tex.) (the "Texas Action"). Defendant acknowledged that  
23 its products "share[d] technical similarities with products ... [accused] of infringement in the  
24 Texas Action." *eBioscience Corp. v. Invitrogen Corp.*, 08-cv-1729 JAH (LSP) (S.D. Cal.), DN 1,  
25 Compl. ¶ 13. In the Texas Action, a Consent Judgment and Permanent Injunction was entered  
26 holding that the accused products in the Texas Action infringed the asserted claims of the '551,  
27 '723, and '069 Patents. *Invitrogen Corp. v. Evident Tech., Inc.*, 08-cv-163-LED/JDL (E.D. Tex.),

1 DN 98, Consent J. ¶ 7. The Texas court further held in that Consent Judgment and Permanent  
2 Injunction that all of the claims of the '551, '723, and '069 Patents are valid and enforceable "in all  
3 respects." *Invitrogen Corp. v. Evident Tech., Inc.*, 08-cv-163-LED/JDL (E.D. Tex.), DN 98,  
4 Consent J. ¶¶ 8-9. Plaintiffs are informed and believe, and on that basis allege, that Defendant is  
5 aware of the outcome in *Invitrogen Corp. v. Evident Tech., Inc.*. Thus, Defendant's infringement  
6 of the '551, '723, and '069 Patents has been and is willful, knowing, and deliberate, entitling  
7 Plaintiffs to increased damages under 35 U.S.C. § 284 and to attorneys' fees and costs incurred in  
8 prosecuting this action under 35 U.S.C. § 285.

9 17. Plaintiffs have been damaged and have suffered irreparable injury due to the  
10 Defendant's acts of infringement, and Plaintiffs will continue to suffer irreparable injury unless  
11 Defendant's acts are enjoined.

12 18. Plaintiffs Life Technologies and Molecular Probes have suffered and will continue  
13 to suffer substantial damage to their business in the form of lost profits by reason of Defendant's  
14 acts of patent infringement as alleged herein, and Plaintiffs Life Technologies and Molecular  
15 Probes are entitled to recover from Defendant the damages sustained as a result of Defendant's  
16 acts.

### 17 **RELIEF REQUESTED**

18 19. Judgment that Defendant has infringed, contributed to the infringement of, and  
19 induced infringement of, literally and/or under the doctrine of equivalents, the asserted claims of  
20 the '551, '723 and '069 Patents;

21 20. That Defendant and its subsidiaries, affiliates, parents, successors, assigns, officers,  
22 agents, servants, employees, attorneys, and all other persons acting in concert or in participation  
23 with it, be temporarily and preliminarily enjoined during the pendency of this action, and  
24 permanently enjoined thereafter, from infringing the '551, '723 and '069 Patents, and specifically  
25 from directly or indirectly making, using, selling, offering for sale, or importing any products or  
26 services embodying the inventions of the '551, '723 and '069 Patents during the life of the claims  
27 of the '551, '723 and '069 Patents without the express written authority of Plaintiffs;

1           21.     That Defendant be directed to fully compensate Plaintiffs for all damages  
2 attributable to Defendant's infringement of the '551, '723 and '069 Patents in an amount  
3 according to proof at trial, but not less than a reasonable royalty;

4           22.     That Defendant be ordered to deliver to Plaintiffs, for destruction at Plaintiffs'  
5 option, all products that infringe the '551, '723 and '069 Patents;

6           23.     That Defendant be required to account for all gains, profits, advantages, and unjust  
7 enrichment derived from its violations of the law;

8           24.     That Defendant's infringement of the '551, '723 and '069 Patents has been willful,  
9 wanton, and intentional;

10          25.     That Plaintiffs be awarded treble damages due to Defendant's willful infringement  
11 of the '551, '723 and '069 Patents;

12          26.     That Plaintiffs be awarded their reasonable attorneys' fees incurred in connection  
13 with this matter;

14          27.     That Plaintiffs be awarded the costs of suit, and an assessment of interest; and,

15          28.     That Plaintiffs have such other, further, and different relief as the evidence may  
16 require and as the Court deems proper under the circumstances.

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1 Dated: April 12, 2011  
2

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22 THE UNIVERSITY OF CALIFORNIA  
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**JURY DEMAND**

Plaintiffs hereby demand a jury trial on all claims, causes of action, issues, and defenses properly triable before a jury

Dated: April 12, 2011

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THE UNIVERSITY OF CALIFORNIA



# **EXHIBIT 1**



US006423551B1

(12) **United States Patent**  
Weiss et al.

(10) **Patent No.:** US 6,423,551 B1  
(45) **Date of Patent:** Jul. 23, 2002

(54) **ORGANO LUMINESCENT SEMICONDUCTOR NANOCRYSTAL PROBES FOR BIOLOGICAL APPLICATIONS AND PROCESS FOR MAKING AND USING SUCH PROBES**

5,537,000 A	7/1996	Alivisatos et al. ....	313/506
5,585,640 A	12/1996	Huston et al. ....	250/483.1
5,674,698 A	10/1997	Zarling et al. ....	435/7.92
5,736,330 A	4/1998	Fulton .....	435/6
5,751,018 A	5/1998	Alivisatos et al. ....	257/64

(75) Inventors: **Shimon Weiss**, Pinole; **Marcel Bruchez, Jr.**, Albany; **Paul Alivisatos**, Oakland, all of CA (US)

**FOREIGN PATENT DOCUMENTS**

EP	0 990 903	4/2000	.....	G01N/33/58
WO	WO 98/04740	2/1998	.....	C12Q/1/68
WO	WO 99/19515	4/1999	.....	C12Q/1/68

(73) Assignee: **The Regents of the University of California**, Oakland, CA (US)

**OTHER PUBLICATIONS**

(\* ) Notice: Subject to any disclaimer, the term of this patent is extended or adjusted under 35 U.S.C. 154(b) by 0 days.

Dabbousi, B.O., et al., "(CdSe) ZnS Core-Shell Quantum Dots: Synthesis and Characterization of a Size Series of Highly Luminescent Nanocrystallites", *Journal of Physical Chemistry B*, vol. 101, 1997, pp. 9463-9475.

(21) Appl. No.: **09/349,833**

(List continued on next page.)

(22) Filed: **Jul. 8, 1999**

**Related U.S. Application Data**

(63) Continuation of application No. 08/978,450, filed on Nov. 25, 1997, now Pat. No. 5,990,479.

(51) **Int. Cl.<sup>7</sup>** ..... **G01N 33/543**

(52) **U.S. Cl.** ..... **436/518**; 424/9.32; 424/9.34; 424/9.341; 424/9.36; 428/402; 428/402.24; 428/403; 428/404; 428/405; 428/406; 436/172; 436/173; 436/524; 436/525; 436/527

(58) **Field of Search** ..... 424/9.32, 9.34, 424/9.341, 9.36; 428/402, 402.24, 403, 404, 405, 406; 436/518, 524, 525, 527, 173, 172

(56) **References Cited**

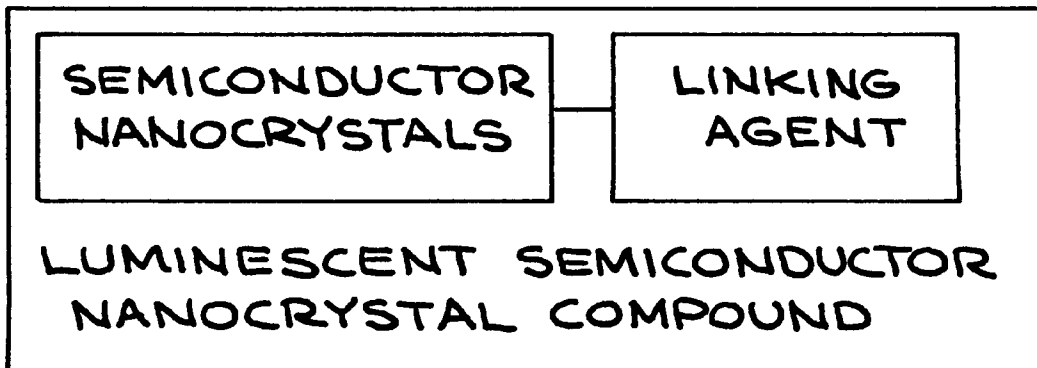
**U.S. PATENT DOCUMENTS**

3,996,345 A	12/1976	Ullman et al. ....	424/12
4,637,988 A	1/1987	Hinshaw et al. ....	436/546
4,777,128 A	10/1988	Lippa .....	435/5
5,262,357 A	11/1993	Alivisatos et al. ....	437/233
5,319,209 A	6/1994	Miyakawa et al. ....	250/459.1
5,460,831 A	* 10/1995	Kossovsky et al. ....	424/493
5,505,928 A	4/1996	Alivisatos et al. ....	423/299

(57) **ABSTRACT**

A semiconductor nanocrystal compound is described capable of linking to an affinity molecule. The compound comprises (1) a semiconductor nanocrystal capable of emitting electromagnetic radiation and/or absorbing energy, and/or scattering or diffracting electromagnetic radiation—when excited by an electromagnetic radiation source or a particle beam; and (2) at least one linking agent, having a first portion linked to the semiconductor nanocrystal and a second portion capable of linking to an affinity molecule. The compound is linked to an affinity molecule to form a semiconductor nanocrystal probe capable of bonding with a detectable substance. Subsequent exposure to excitation energy will excite the semiconductor nanocrystal in the probe, causing the emission of electromagnetic radiation. Further described are processes for respectively: making the semiconductor nanocrystal compound; making the semiconductor nanocrystal probe; and using the probe to determine the presence of a detectable substance in a material.

**26 Claims, 3 Drawing Sheets-**



## US 6,423,551 B1

Page 2

## OTHER PUBLICATIONS

- Peng, Xiaogang, et al., "Epitaxial Growth of Highly Luminescent CdSe/CdS Core/Shell Nanocrystals with Photostability and Electronic Accessibility", *Journal of the American Chemical Society*, vol. 119, No. 30, pp. 7019–7029.
- Alivisatos, A. P., "Semiconductor Clusters, Nanocrystals, and Quantum Dots," *Science* 271 (Feb. 16, 1996):933–937.
- Alivisatos, A. P., "Perspectives on the Physical Chemistry of Semiconductor Nanocrystals," *J. Phys. Chem.* 100 (1996):13226–13239.
- Alivisatos, A. Paul, et al., "Organization of 'Nanocrystal Molecules' Using DNA," *Nature* 382 (Aug. 15, 1996):609–611.
- Beverloo, H.B., et al., "Preparation and Microscopic Visualization of Multicolor Luminescent Immunophosphors," Chapter 4 of Beverloo, H.B., "Inorganic Crystals as Luminescent Labels: Their Applications in Immunocytochemistry and Time-Resolved Microscopy," Ph.D. dissertation, University of Leiden (The Netherlands), May 13, 1992, pp. 553–573.
- Bruchez, Marcel P., Jr., "Luminescent Semiconductor Nanocrystals: Intermittent Behavior and Use as Fluorescent Biological Probes," Ph.D. dissertation, University of California, Dec. 17, 1998.
- Bruchez, Marcel, Jr., et al., "Semiconductor Nanocrystals as Fluorescent Probes for Biology", *Cytometry Supp.* 9 (1998):26.
- Chan, Warren C.W., et al., "Quantum Dot Bioconjugates for Ultrasensitive Nonisotopic Detection," *Science* 281 (Sep. 25, 1998):2016–2018.
- Coffer, Jeffrey L., et al., "Characterization of Quantum-Confined CdS Nanocrystallites Stabilized by Deoxyribonucleic Acid (DNA)," *Nanotechnol.* 3 (1992):69–76.
- Cook, Neil D., "Scintillation Proximity Assay: A Versatile High-Throughput Screening Technology," *Drug Discovery Today* 1 (Jul. 1996):287–294.
- Correa-Duarte, Miguel A., et al., "Stabilization of CdS Semiconductor Nanoparticles Against Photodegradation by Silica Coating Procedure," *Chem. Phys. Lett.* 286 (Apr. 17, 1998):497–501.
- Jacoby, Mitch, "Quantum Dots Meet Biomolecules," *C&E News* 76 (Sep. 28, 1998):Copied from the Internet as pp. 1–3.
- Kagan, C.R., et al., "Electronic Energy Transfer in CdSe Quantum Dot Solids," *Phys. Rev. Lett.* 76 (Feb. 26, 1996):1517–1520.
- Leff, David N., "Color-Coding Quantum Dots Debut with Promising Careers in Clinical Diagnostics Field," *Bioworld Today*, Sep. 25, 1998, Copied from the Internet as pp. 1–2.
- Liz-Marzán, Luis M., et al., "Synthesis of Nanosized Gold-Silica Core-Shell Particles," *Langmuir* 12 (1996):4329–4335.
- Mahtab, Rahina, et al., "Preferential Adsorption of a 'Kinked' DNA to a Neutral Curved Surface: Comparisons to and Implications for Nonspecific DNA-Protein Interactions," *J. Am. Chem. Soc.* 118 (1996):7028–7032.
- Mahtab, Rahina, et al., "Protein-Sized Quantum Dot Luminescence Can Distinguish Between 'Straight,' 'Bent,' and 'Kinked' Oligonucleotides," *J. Am. Chem. Soc.* 117 (1995):9099–9100.
- Murphy, Catherine J., et al., "Quantum Dots as Inorganic DNA-Binding Proteins," *Mat. Res. Soc. Symp. Proc.* 452 (1997):597–600.
- Peng, Xiaogang, et al., "Synthesis and Isolation of a Homodimer of Cadmium Selenide Nanocrystals," *Angewandte Chemie-International Edition in English*, 36 (1997):145–147.
- Service, Robert F., "Semiconductor Beacons Light Up Cell Structures," *Science* 281 (Sep. 25, 1998):1930–1931.
- Shröck, E., et al., "Multicolor Spectral Karyotyping of Human Chromosomes," *Science* 273 (Jul. 26, 1996):494–497.
- Zhang, Yu-zhong, et al., "Novel Flow Cytometry Compensation Standards: Internally Stained Fluorescent Microspheres with Matched Emission Spectra and Long-Term Stability," *Cytometry* 33 (1998):244–248.
- Lacoste, T.D., et al., "Super Resolution Molecular Ruler Using Single Quantum Dots", *Biophysical Journal*, vol. 78, Jan. 2000, p. 402A, XP-000933548 Abstract.
- Bruchez, Marcel, Jr., et al., "Semiconductor Nanocrystals as Fluorescent Biological Labels", *Science*, vol. 281, Sep. 25, 1998, pp. 2013–2016.

\* cited by examiner

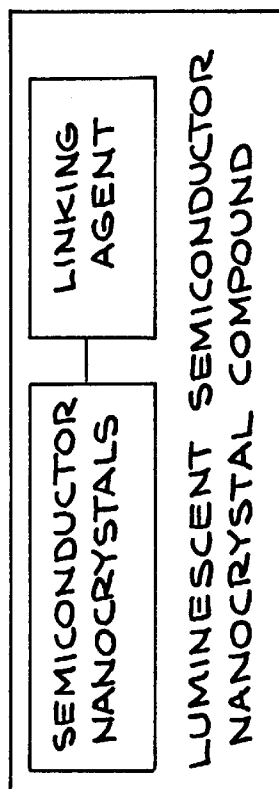


FIG. 1

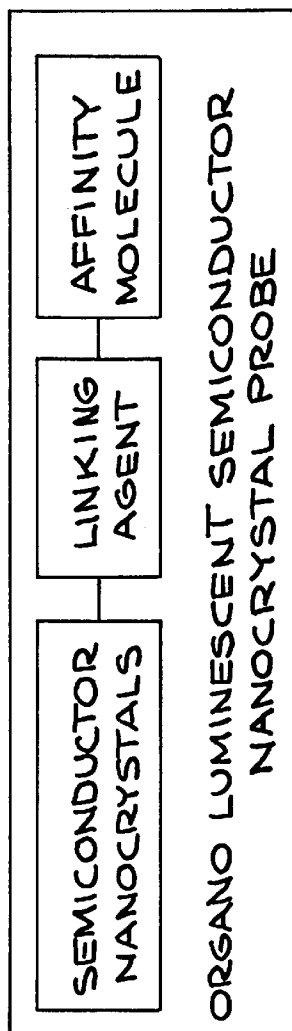


FIG. 2

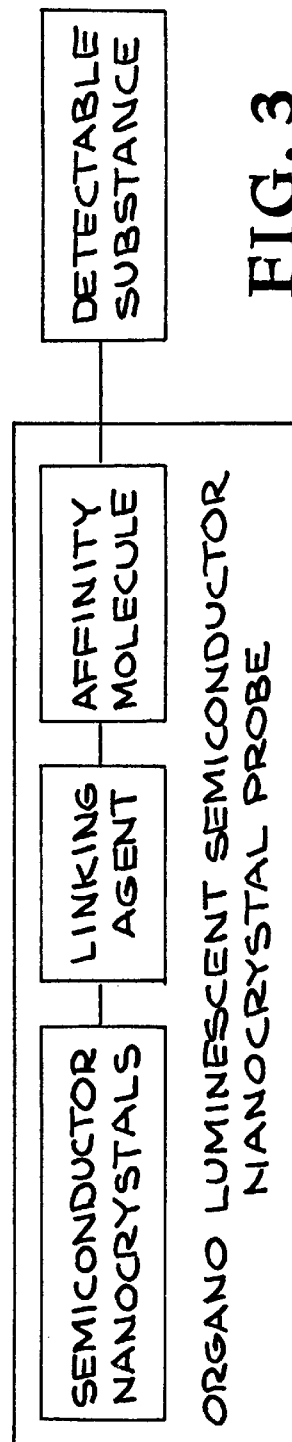
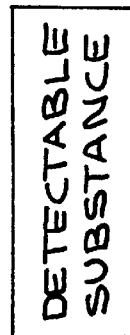


FIG. 3



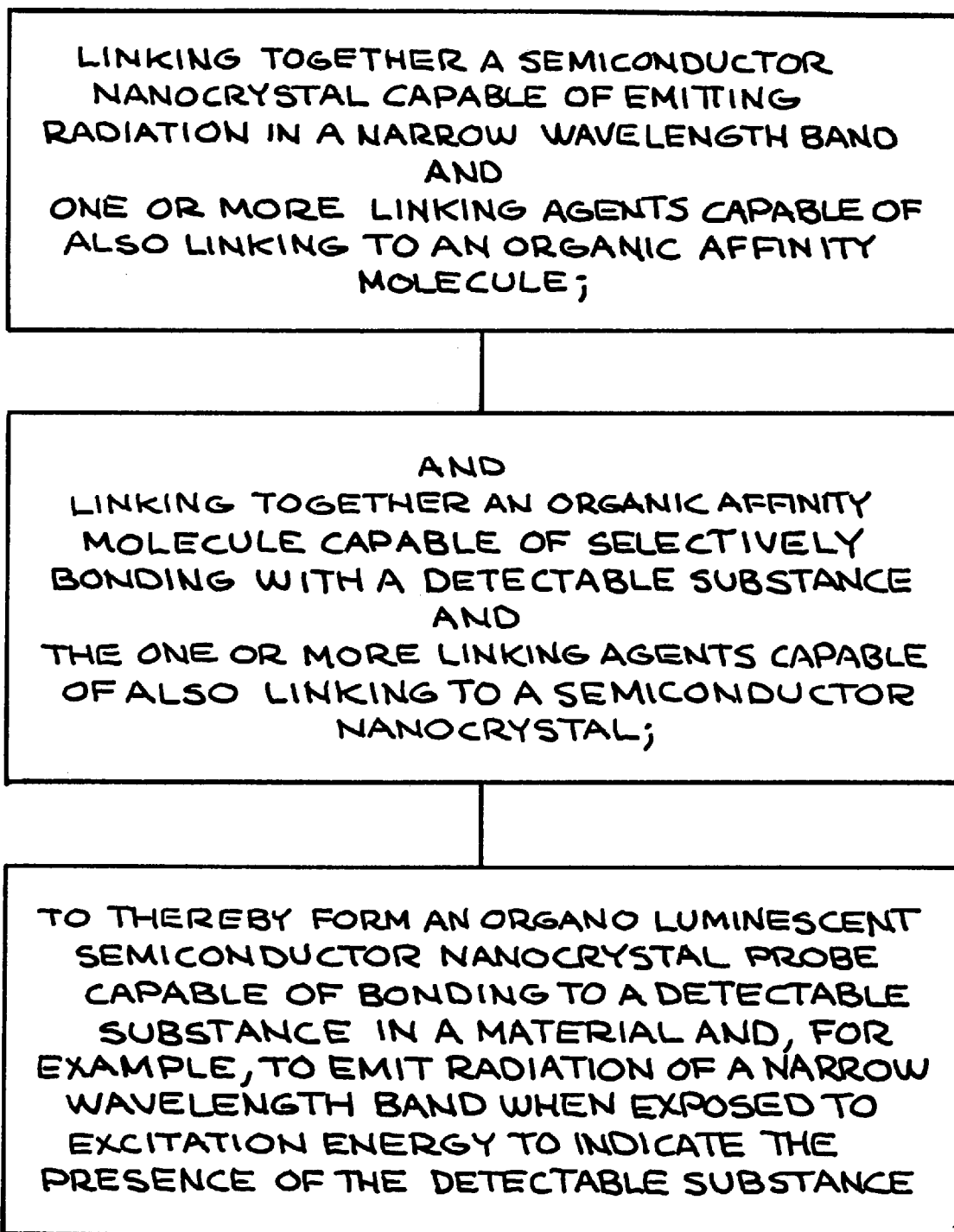


FIG. 4

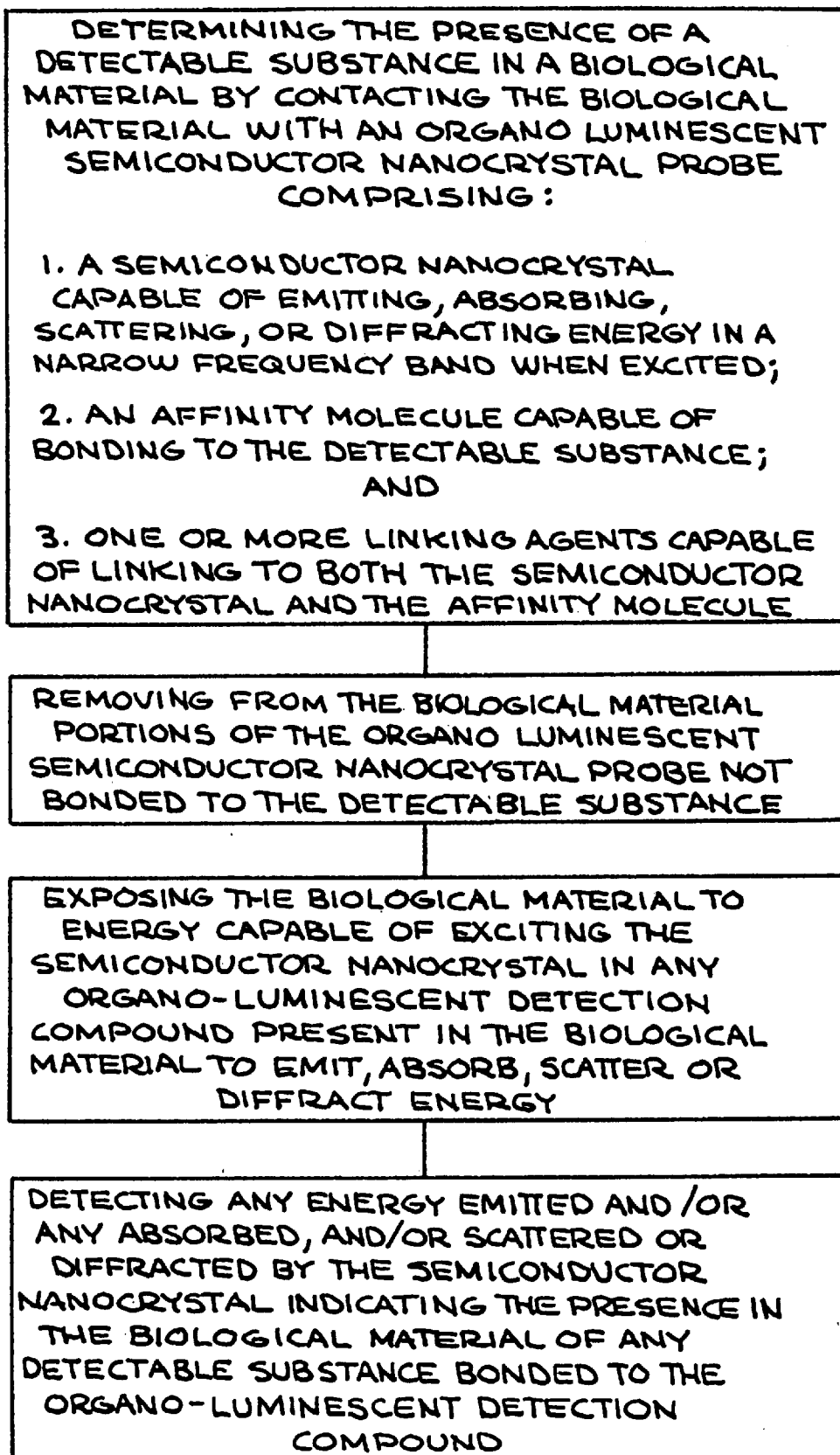


FIG. 5

US 6,423,551 B1

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**ORGANO LUMINESCENT  
SEMICONDUCTOR NANOCRYSTAL PROBES  
FOR BIOLOGICAL APPLICATIONS AND  
PROCESS FOR MAKING AND USING SUCH  
PROBES**

CROSS-REFERENCE TO RELATED  
APPLICATION

This application is a continuation of U.S. patent application Ser. No. 08/978,450 filed Nov. 25, 1997, and now issued as U.S. Pat. No. 5,990,479 on Nov. 23, 1999.

The invention described herein arose in the course of, or under, Contract No. DE-AC03-SF00098 between the United States Department of Energy and the University of California for the operation of the Ernest Orlando Lawrence Berkeley National Laboratory. The Government may have rights to the invention.

BACKGROUND OF THE INVENTION

1. Field of the Invention

This invention relates to organo luminescent semiconductor nanocrystal probes for biological applications wherein the probes includes a plurality of semiconductor nanocrystals capable of luminescence and/or absorption and/or scattering or diffraction when excited by a radiation or particle beam.

2. Description of the Related Art

Fluorescent labeling of biological systems is a well known analytical tool used in modern biotechnology as well as analytical chemistry. Applications for such fluorescent labeling include technologies such as medical (and non-medical) fluorescence microscopy, histology, flow cytometry, fluorescence in-situ hybridization (medical assays and research), DNA sequencing, immuno-assays, binding assays, separation, etc.

Conventionally, such fluorescent labeling involves the use of an organic dye molecule bonded to a moiety which, in turn, selectively bonds to a particular biological system, the presence of which is then identified by excitation of the dye molecule to cause it to fluoresce. There are a number of problems with such an analytical system. In the first place, the emission of light of visible wavelengths from an excited dye molecule usually is characterized by the presence of a broad emission spectrum as well as a broad tail of emissions on the red side of the spectrum, i.e., the entire emission spectrum is rather broad. As a result, there is a severe limitation on the number of different color organic dye molecules which may be utilized simultaneously or sequentially in an analysis since it is difficult to either simultaneously or even non-simultaneously detect or discriminate between the presence of a number of different detectable substances due to the broad spectrum emissions and emission tails of the labelling molecules. Another problem is that most dye molecules have a relatively narrow absorption spectrum, thus requiring either multiple excitation beams used either in tandem or sequentially for multiple wavelength probes, or else a broad spectrum excitation source which is sequentially used with different filters for sequential excitation of a series of probes respectively excited at different wavelengths.

Another problem frequently encountered with existing dye molecule labels is that of photostability. Available fluorescent molecules bleach, or irreversibly cease to emit light, under repeated excitation ( $10^4$ - $10^8$ ) cycles of absorption/emission. These problems are often surmounted

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by minimizing the amount of time that the sample is exposed to light, and by removing oxygen and/or other radical species from the sample.

In addition, the probe tools used for the study of these systems by electron microscopy techniques are completely different from the probes used for study by fluorescence. Thus, it is not possible to label a material with a single type of probe for both electron microscopy and for fluorescence.

It would, therefore, be desirable to provide a stable probe material for biological applications having a wide absorption band and capable of exhibiting either a detectable change in absorption or of emitting radiation in a narrow wavelength band, without the presence of the large red emission tails characteristic of dye molecules (thereby permitting the simultaneous use of a number of such probe materials, each emitting light of a different narrow wavelength band) and/or capable of scattering or diffracting radiation. It would also be equally desirable to provide a single, stable probe material which can be used to image the same sample by both light and electron microscopy.

SUMMARY OF THE INVENTION

The invention comprises a luminescent semiconductor nanocrystal compound capable of linking to an affinity molecule to form an organo luminescent semiconductor nanocrystal probe capable of luminescence and/or absorption and/or scattering or diffracting when excited by an electromagnetic radiation source (of broad or narrow bandwidth) or a particle beam, and capable of exhibiting a detectable change in absorption and/or of emitting radiation in a narrow wavelength band and/or scattering or diffracting when so excited. The luminescent semiconductor nanocrystal compound preferably comprises: (1) a semiconductor nanocrystal capable of luminescence and/or absorption and/or scattering or diffraction when excited by an electromagnetic radiation source (of broad or narrow bandwidth) or a particle beam, and capable of exhibiting a detectable change in absorption and/or of emitting radiation in a narrow wavelength band and/or scattering or diffracting when excited; and (2) a linking agent having a first portion linked to the semiconductor nanocrystal, and a second portion capable of linking to an affinity molecule.

The invention further comprises an organo luminescent semiconductor nanocrystal probe formed by linking the above described luminescent semiconductor nanocrystal compound to an affinity molecule capable of bonding to a detectable substance in a material. As a result the organo luminescent semiconductor nanocrystal probe, in one embodiment, is capable of absorbing or scattering or diffracting energy from either a particle beam or an electromagnetic radiation source (of broad or narrow bandwidth), and is capable of emitting electromagnetic radiation in a narrow wavelength band when so excited; while in another embodiment the amount of energy so absorbed, or scattered, or diffracted from either a particle beam or an electromagnetic radiation source (of broad or narrow bandwidth), is detectable, i.e., the change in absorption, scattering, or diffraction is detectable.

Therefore, treatment of a material with the organo luminescent semiconductor nanocrystal probe, and subsequent exposure of this treated material to excitation energy (from either a particle beam or an electromagnetic radiation source of broad or narrow bandwidth) to determine the presence of the detectable substance within the material, will excite the semiconductor nanocrystals in the organo luminescent semiconductor nanocrystal probe bonded to the detectable

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substance, resulting in the emission of electromagnetic radiation of a narrow wavelength band and/or a detectable change in the amount of energy being absorbed and/or scattered or diffracted, signifying the presence, in the material, of the detectable substance bonded to the organo luminescent semiconductor nanocrystal probe.

The invention also comprises a process for making the luminescent semiconductor nanocrystal compound and for making the organo luminescent semiconductor nanocrystal probe comprising the luminescent semiconductor nanocrystal compound linked to an affinity molecule capable of bonding to a detectable substance. The organo luminescent semiconductor nanocrystal probe of the invention is stable with respect to repeated excitation by light, or exposure to oxygen or other radicals. The invention further comprises a process for treating a material, such as a biological material, to determine the presence of a detectable substance in the material which comprises contacting the material with the organo luminescent semiconductor nanocrystal probe, removing from the material portions of the organo luminescent semiconductor nanocrystal probe not bonded to the detectable substance, and then exposing the material to activation energy from either an electromagnetic radiation source (of broad or narrow bandwidth) or a particle beam. The presence of the detectable substance in the material is then determined either by measuring the absorption of energy by the organo luminescent semiconductor nanocrystal probe and/or detecting the emission of radiation of a narrow wavelength band by the organo luminescent semiconductor nanocrystal probe and/or detecting the scattering or diffraction by the organo luminescent semiconductor nanocrystal probe, indicative (in either case) of the presence of the organo luminescent semiconductor nanocrystal probe bonded to the detectable substance in the material.

#### BRIEF DESCRIPTION OF THE DRAWINGS

FIG. 1 is a block diagram of the luminescent semiconductor nanocrystal compound of the invention.

FIG. 2 is a block diagram of the organo luminescent semiconductor nanocrystal probe of the invention.

FIG. 3 is a block diagram showing the affinity between a detectable substance and the organo luminescent semiconductor nanocrystal probe of the invention.

FIG. 4 is a flow sheet illustrating the process of forming the organo luminescent semiconductor nanocrystal probe of the invention.

FIG. 5 is a flow sheet illustrating a typical use of the organo luminescent semiconductor nanocrystal probe of the invention in detecting the presence of a detectable substance in a material such as a biological material.

#### DETAILED DESCRIPTION OF THE INVENTION

The invention comprises a luminescent semiconductor nanocrystal compound capable of linking to an organic molecule and capable of exhibiting a detectable change in absorption and/or of emitting electromagnetic radiation in a narrow wavelength band and/or scattering or diffracting when excited by either an electromagnetic radiation source (of broad or narrow bandwidth) or a particle beam. The luminescent semiconductor nanocrystal compound, in turn, comprises: (1) semiconductor nanocrystals capable of exhibiting a detectable change in absorption and/or of emitting electromagnetic radiation in a narrow wavelength band when excited by either an electromagnetic radiation source

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(of broad or narrow bandwidth) or a particle beam; and (2) one or more linking agents each having a first portion linked to the semiconductor nanocrystal and a second portion capable of linking to an organic affinity molecule.

The invention also comprises the above described luminescent semiconductor nanocrystal compound linked to the organic affinity molecule (through the linking agent) to form an organo luminescent semiconductor nanocrystal probe capable of bonding to a detectable substance and capable of exhibiting a detectable change in absorption and/or of emitting electromagnetic radiation in a narrow wavelength band and/or scattering or diffracting when excited by either an electromagnetic radiation source (of broad or narrow bandwidth) or a particle beam. Treatment of a material (typically a biological material) with the organo luminescent semiconductor nanocrystal probe, and subsequent exposure of this treated material to excitation energy, as described above, to determine the presence of the detectable substance within the material, will excite the semiconductor nanocrystal in the organo luminescent semiconductor nanocrystal probe bonded to the detectable substance, causing the detectable change in absorption and/or emission of electromagnetic radiation of a narrow wavelength band and/or scattering or diffraction signifying (in either instance) the presence in the material, of the detectable substance bonded to the organo luminescent semiconductor nanocrystal probe.

The invention also comprises a process for making the luminescent semiconductor nanocrystal compound, and a process for making the organo luminescent semiconductor nanocrystal probe comprising the luminescent semiconductor nanocrystal compound linked to an affinity molecule capable of bonding to a detectable substance.

The invention further comprises a process for treating a material, such as a biological material, to determine the presence of a detectable substance in the material which comprises: (1) contacting the material with the organo luminescent semiconductor nanocrystal probe, (2) removing from the material portions of the organo luminescent semiconductor nanocrystal probe not bonded to the detectable substance, (3) exposing the material to energy (such as the above-described electromagnetic energy source or particle beam) capable of exciting the semiconductor nanocrystal to cause a detectable change in absorption and/or emission of electromagnetic radiation of a narrow wavelength band and/or scattering or diffraction signifying (in either instance) the presence of the organo luminescent semiconductor nanocrystal probe bonded to the detectable substance in the material, and (4) detecting either the change in absorbed energy or the electromagnetic radiation emitted or the scattering or diffraction by the semiconductor nanocrystal in the organo luminescent semiconductor nanocrystal probe.

#### a. Definitions

By use of the terms "nanometer crystal" or "nanocrystal" herein is meant an organic or inorganic single crystal particle having an average cross-section no larger than about 20 nanometers (nm) or  $20 \times 10^{-9}$  meters (200 Angstroms), preferably no larger than about 10 nm (100 Angstroms) and a minimum average cross-section of about 1 nm, although in some instances a smaller average cross-section nanocrystal, i.e., down to about 0.5 nm (5 Angstroms), may be acceptable. Typically the nanocrystal will have an average cross-section ranging in size from about 1 nm (10 Angstroms) to about 10 nm (100 angstroms).

By use of the term "semiconductor nanocrystal" is meant a nanometer crystal or nanocrystal of Group II-VI and



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Group III–V semiconductor compounds capable of emitting electromagnetic radiation upon excitation, although the use of Group IV semiconductors such as germanium or silicon, or the use of organic semiconductors, may be feasible under certain conditions.

By use of the term “a narrow wavelength band”, with regard to the electromagnetic radiation emission of the semiconductor nanocrystal, is meant a wavelength band of emissions not exceeding about 40 nm, and preferably not exceeding about 20 nm in width and symmetric about the center, in contrast to the emission bandwidth of about 100 nm for a typical dye molecule, with a red tail which may extend the band width out as much as another 100 nm. It should be noted that the bandwidths referred to are determined from measurement of the width of the emissions at half peak height (FWHM), and are appropriate in the range of 200 nm to 2000 nm.

By use of the term “a broad absorption band”, with regard to the electromagnetic radiation absorption of the semiconductor nanocrystal is meant a continuously increasing absorption from the onset, which occurs near to, but at slightly higher energy than the “narrow wavelength band” of the emission. This is in contrast to the “narrow absorption band” of dye molecules which occurs near the emission peak on the high energy side, but drops off rapidly away from that wavelength.

By use of the term “detectable substance” is meant an entity or group, the presence or absence of which in a material such as a biological material, is to be ascertained by use of the organo-luminescent semiconductor nanocrystal probe of the invention.

By use of the term “affinity molecule” is meant the portion of the organo luminescent semiconductor nanocrystal probe of the invention which will selectively bond to a detectable substance (if present) in the material (e.g., biological material) being analyzed.

By use of the term “linking agent” is meant a substance capable of linking with a semiconductor nanocrystal and also capable of linking to an affinity molecule.

The terms “link” and “linking” are meant to describe the adherence between the affinity molecule and the semiconductor nanocrystals, either directly or through a moiety identified herein as a linking agent. The adherence may comprise any sort of bond, including, but not limited to, covalent, ionic, hydrogen bonding, Van der Waals’ forces, or mechanical bonding, etc.

The terms “bond” and “bonding” are meant to describe the adherence between the affinity molecule and the detectable substance. The adherence may comprise any sort of bond, including, but not limited to, covalent, ionic, or hydrogen bonding, Van der Waals’ forces, or mechanical bonding, etc.

The term “luminescent semiconductor nanocrystal compound”, as used herein, is intended to define a semiconductor nanocrystal linked to one or more linking agents and capable of linking to an affinity molecule, while the term “organo-luminescent semiconductor nanocrystal probe” is intended to define a luminescent semiconductor nanocrystal compound linked to an affinity molecule.

The term “glass” as used herein is intended to include one or more oxides of silicon, boron, and/or phosphorus, or a mixture thereof, as well as the further optional inclusion of one or more metal silicates, metal borates or metal phosphates therein.

#### b. The Semiconductor Nanocrystals

The semiconductor nanocrystals useful in the practice of the invention include nanocrystals of Group II–VI semicon-

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ductors such as MgS, MgSe, MgTe, CaS, CaSe, CaTe, SrS, SrSe, SrTe, BaS, BaSe, BaTe, ZnS, ZnSe, ZnTe, CdS, CdSe, CdTe, HgS, HgSe, and HgTe; and nanocrystals of Group III–V semiconductors such as GaAs, InGaAs, InP, and InAs.

As mentioned above, the use of Group IV semiconductors such as germanium or silicon, or the use of organic semiconductors, may also be feasible under certain conditions.

Formation of nanometer crystals of Group III–V semiconductors is described in copending and commonly assigned Alivisatos et al. U.S. Pat. No. 5,751,018; Alivisatos et al. U.S. Pat. No. 5,505,928; and Alivisatos et al. U.S. Pat. No. 5,262,357, which also describes the formation of Group II–VI semiconductor nanocrystals, and which is also assigned to the assignee of this invention. Also described therein is the control of the size of the semiconductor nanocrystals during formation using crystal growth terminators. The teachings of Alivisatos et al. U.S. Pat. No. 5,751,018, and Alivisatos et al. U.S. Pat. No. 5,262,357 are each hereby specifically incorporated by reference.

In a preferred embodiment, the nanocrystals are used in a core/shell configuration wherein a first semiconductor nanocrystal forms a core ranging in diameter, for example, from about 20 Å to about 100 Å, with a shell of another semiconductor nanocrystal material grown over the core nanocrystal to a thickness of, for example, 1–10 monolayers in thickness. When, for example, a 1–10 monolayer thick shell of CdS is epitaxially grown over a core of CdSe, there is a dramatic increase in the room temperature photoluminescence quantum yield. Formation of such core/shell nanocrystals is described more fully in a publication by one of us with others entitled “Epitaxial Growth of Highly Luminescent CdSe/CdS Core/Shell Nanocrystals with Photostability and Electronic Accessibility”, by Peng, Schlamp, Kadavanich, and Alivisatos, published in the Journal of the American Chemical Society, Volume 119, No. 30, 1997, at pages 7019–7029, the subject matter of which is hereby specifically incorporated herein by reference.

The semiconductor nanocrystals used in the invention will have a capability of emitting light within a narrow wavelength band of about 40 nm or less, preferably about 20 nm or less, thus permitting the simultaneous use of a plurality of differently colored organo luminescent semiconductor nanocrystal probes with different semiconductor nanocrystals without overlap (or with a small amount of overlap) in wavelengths of emitted light (unlike the use of dye molecules with broad emission lines (e.g., ~100 nm) and broad tails of emission (e.g., another 100 nm) on the red side of the spectrum), thus allowing for the simultaneous detection of a plurality of detectable substances.

#### c. Affinity Molecule

The particular affinity molecule forming a part of the organo-luminescent semiconductor nanocrystal probe of the invention will be selected based on its affinity for the particular detectable substance whose presence or absence, for example, in a biological material, is to be ascertained. Basically, the affinity molecule may comprise any molecule capable of being linked to a luminescent semiconductor nanocrystal compound which is also capable of specific recognition of a particular detectable substance. In general, any affinity molecule useful in the prior art in combination with a dye molecule to provide specific recognition of a detectable substance will find utility in the formation of the organo-luminescent semiconductor nanocrystal probes of the invention. Such affinity molecules include, by way of

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example only, such classes of substances as monoclonal and polyclonal antibodies, nucleic acids (both monomeric and oligomeric), proteins, polysaccharides, and small molecules such as sugars, peptides, drugs, and ligands. Lists of such affinity molecules are available in the published literature such as, by way of example, the "Handbook of Fluorescent Probes and Research Chemicals", (sixth edition) by R. P. Haugland, available from Molecular Probes, Inc.

#### d. The Linking Agent

The organo-luminescent semiconductor nanocrystal probe of the invention will usually find utility with respect to the detection of one or more detectable substances in organic materials, and in particular to the detection of one or more detectable substances in biological materials. This requires the presence, in the organo-luminescent semiconductor nanocrystal probe, of an affinity molecule or moiety, as described above, which will bond the organo-luminescent semiconductor nanocrystal probe to the detectable substance in the organic/biological material so that the presence of the detectable material may be subsequently ascertained. However, since the semiconductor nanocrystals are inorganic, they may not bond directly to the organic affinity molecule. In these case therefore, there must be some type of linking agent present in the organo-luminescent semiconductor nanocrystal probe which is capable of forming a link to the inorganic semiconductor nanocrystal as well as to the organic affinity molecule in the organo-luminescent semiconductor nanocrystal probe.

One form in which the semiconductor nanocrystal may be linked to an affinity molecule via a linking agent is by coating the semiconductor nanocrystal with a thin layer of glass, such as silica ( $\text{SiO}_x$  where  $x=1-2$ ), using a linking agent such as a substituted silane, e.g., 3-mercaptopropyl-trimethoxy silane to link the nanocrystal to the glass. The glass-coated semiconductor nanocrystal may then be further treated with a linking agent, e.g., an amine such as 3-aminopropyl-trimethoxysilane, which will function to link the glass-coated semiconductor nanocrystal to the affinity molecule. That is, the glass-coated semiconductor nanocrystal may then be linked to the affinity molecule. It is within

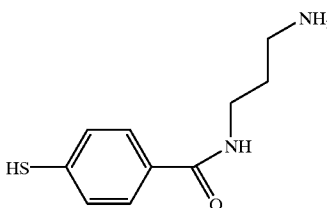
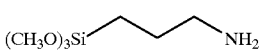
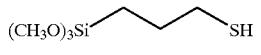
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the contemplation of this invention that the original luminescent semiconductor nanocrystal compound may also be chemically modified after it has been made in order to link effectively to the affinity molecule. A variety of references summarize the standard classes of chemistry which may be used to this end, in particular the "Handbook of Fluorescent Probes and Research Chemicals", (6th edition) by R. P. Haugland, available from Molecular Probes, Inc., and the book "Bioconjugate Techniques", by Greg Hermanson, available from Academic Press, New York.

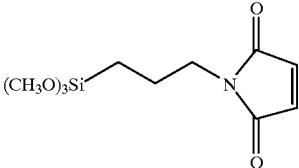
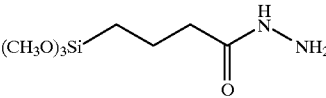
When the semiconductor nanocrystal is coated with a thin layer of glass, the glass, by way of example, may comprise a silica glass ( $\text{SiO}_x$  where  $x=1-2$ ), having a thickness ranging from about 0.5 nm to about 10 nm, and preferably from about 0.5 nm to about 2 nm.

The semiconductor nanocrystal is coated with the coating of thin glass, such as silica, by first coating the nanocrystals with a surfactant such as tris-octyl-phosphine oxide, and then dissolving the surfactant-coated nanocrystals in a basic methanol solution of a linking agent, such as 3-mercaptopropyl-tri-methoxy silane, followed by partial hydrolysis which is followed by addition of a glass-affinity molecule linking agent such as amino-propyl trimethoxysilane which will link to the glass and serve to form a link with the affinity molecule. When the linking agent does not involve the use of a glass coating on the semiconductor nanocrystal, it may comprise a number of different materials, depending upon the particular affinity molecule, which, in turn, depends upon the type of detectable material being analyzed for. It should also be noted that while an individual linking agent may be used to link to an individual semiconductor nanocrystal, it is also within the contemplation of the invention that more than one linking agent may bond to the same semiconductor nanocrystal and vice versa.

A few examples of the types of linking agents which may be used to link to both the semiconductor nanocrystal (or to a glass coating on the nanocrystal) and to the organic affinity molecule in the probe are illustrated in the table below, it being understood that this is not intended to be an exhaustive list:

Linking Agent	
Structure	Name
	N-(3-aminopropyl)3-mercapto-benzamide
	3-aminopropyl-trimethoxysilane
	3-mercaptopropyl-trimethoxysilane

-continued

<u>Linking Agent</u>	
Structure	Name
	3-maleimidopropyl-trimethoxysilane
	3-hydrazidopropyl-trimethoxysilane

It should be further noted that a plurality of polymerizable linking agents may be used together to form an encapsulating net or linkage around an individual nanocrystal (or group of nanocrystals). This is of particular interest where the particular linking agent is incapable of forming a strong bond with the nanocrystal. Examples of linking agents capable of bonding together in such a manner to surround the nanocrystal with a network of linking agents include, but are not limited to: diacetylenes, acrylates, acrylamides, vinyl, styryl, and the aforementioned silicon oxide, boron oxide, phosphorus oxide, silicates, borates and phosphates.

#### e. The Excitation of the Probe and Detection of Emission/Absorption

As previously mentioned, the organo luminescent semiconductor nanocrystal probe of the invention is capable of being excited over a broad bandwidth, yet exhibits emission in a narrow wavelength band, in contrast to the dye molecules used in the prior art. Thus electromagnetic radiation of wavelength ranging from x-ray to ultraviolet to visible to infrared waves may be used to excite the luminescent semiconductor nanocrystals in the probe. In addition, the luminescent semiconductor nanocrystals are capable of excitation from bombardment with a particle beam such as an electron beam (e-beam). Furthermore, because of the broad bandwidth at which the luminescent semiconductor nanocrystals are excitable, one may use a common excitation source for the simultaneous excitation of several probes, i.e., several probes which give off radiation at different frequencies, thus permitting simultaneous excitation and detection of the presence of several probes indicating, for example, the presence of several detectable substances in the material being examined.

Thus, for example, a laser radiation source of a given frequency, e.g., blue light, may be used to excite a first organo luminescent semiconductor nanocrystal probe capable of emitting radiation of a second frequency, e.g., red light, indicating the presence, in the material being illuminated, of a first detectable substance to which the particular red light-emitting organo luminescent semiconductor nanocrystal probe has bonded. At the same time, the same blue light laser source may also be exciting a second organo luminescent semiconductor nanocrystal probe (in the same material) capable of emitting radiation of a third frequency, e.g., green light, indicating the presence, in the material being illuminated, of a second detectable substance to which the particular green light-emitting organo lumines-

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cent semiconductor nanocrystal probe has bonded. Thus, unlike the prior art, multiple excitation sources need not be used (because of the broad bandwidth in which the organo luminescent semiconductor nanocrystal probe of the invention is capable of being excited), and the narrow band of emission of the specific semiconductor nanocrystals in each probe makes possible the elimination of sequencing and/or elaborate filtering to detect the emitted radiation.

With respect to the absorption of energy by the probe of the invention, when the excitation source is an electron beam, or an X-ray source, the presence of the organo luminescent semiconductor nanocrystal probe bonded to the detectable substance of interest in the material being analyzed can be ascertained using a commercially available energy absorption or scattering or diffraction detection system wherein changes in absorption or scattering cross section or in diffraction of the material being analyzed can be detected, signifying the presence of the probe in the material, which, in turn, indicates the presence of the detectable substance to which the probe is bonded in the material being analyzed. In addition, it may be possible to use electron or X-ray sources to detect the presence of the organo luminescent semiconductor nanocrystal probe bonded to the detectable substance by using a conventional detection system for the emission of visible light to observe the visible emission in the narrow wavelength of emission of the probe.

The following examples will serve to further illustrate the formation of the organo luminescent semiconductor nanocrystal probes of the invention, as well as their use in detecting the presence of a detectable substance in a material such as a biological material.

#### Example 1

To illustrate the formation of the luminescent semiconductor nanocrystal compound (comprising the semiconductor nanocrystals linked to a linking agent) 20 ml. of a 5 mM solution of (4-mercapto)benzoic acid was prepared with a pH of 10 using  $(\text{CH}_3)_4\text{NOH}\cdot 5\text{H}_2\text{O}$ . 20 mg of tris-octylphosphine oxide coated CdSe/CdS core/shell nanocrystals were added to the solution and stirred until completely dissolved. The resultant nanocrystal/linking agent solution was heated for 5 hours at 50–60° C. and then concentrated to a few ml by evaporation. Then an equal volume of acetone was added and the nanocrystals precipitate out of solution homogeneously. The precipitate was then washed with acetone, dried, and then can be stored.

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The luminescent semiconductor nanocrystal compound prepared above can be linked with an appropriate affinity molecule to form the organo luminescent semiconductor nanocrystal probe of the invention to treat a biological material to determine the presence or absence of a detectable substance. That is, the luminescent semiconductor nanocrystal compound prepared above can be linked, for example, with avidin or streptavidin (as the affinity molecule) to form an organo luminescent semiconductor nanocrystal probe to treat a biological material to ascertain the presence of biotin; or the luminescent semiconductor nanocrystal compound prepared above can be linked with anti-digoxigenin to form an organo luminescent semiconductor nanocrystal probe to treat a biological material to ascertain the presence of digoxigenin.

#### Example 2

To illustrate the formation of luminescent semiconductor nanocrystal compound (comprising glass-coated semiconductor nanocrystals linked to a linking agent), 50  $\mu$ l of 3-mercaptopropyl-trimethoxy silane was added to 40 ml of an anhydrous solution of 25 vol. % dimethylsulfoxide in methanol, and the pH was adjusted to 10–11 using  $(\text{CH}_3)_4\text{NOH}\cdot 5\text{H}_2\text{O}$ . 10 mg of tris-octylphosphine oxide coated CdSe/CdS core-shell particles, prepared by the technique described in the aforementioned Peng, Schlamp, Kadavanich, and Alivisatos article, were then dissolved in this solution, and stirred for several hours. The solution was diluted with 40 ml of methanol adjusted to a pH of 10 with  $(\text{CH}_3)_4\text{NOH}\cdot 5\text{H}_2\text{O}$ , and heated for 1 hour at 69° C. The solution was stirred for an hour, and 40 ml of a 90 vol. % methanol/9.89 vol. %  $\text{H}_2\text{O}$ /0.1 vol. % trimethoxysilylpropyl urea/0.01 vol. % aminopropyl-trimethoxy silane solution which had been stirring for at least an hour, was added, and stirred for 2 hours. Subsequently the reaction was heated to 69° C. for 15 minutes, and then cooled. 10 ml of a 10 vol. % chlorotrimethyl silane solution in methanol which had been adjusted to a pH of 10 using  $(\text{CH}_3)_4\text{NOH}\cdot 5\text{H}_2\text{O}$  was added, stirred for 2 hours, then heated to 60° C., and then partially concentrated under vacuum. Once the methanol had all evaporated, the solution was precipitated with acetone as an oil product comprising the luminescent semiconductor nanocrystal compound. The luminescent semiconductor nanocrystal compound may then be redissolved in water, and in a variety of buffer solutions to prepare it for linking it to an affinity molecule to form the organo luminescent semiconductor nanocrystal probe of the invention to treat a biological material to determine the presence or absence of a detectable substance.

Thus, the invention provides an organo luminescent semiconductor nanocrystal probe containing a semiconductor nanocrystal capable, upon excitation by either electromagnetic radiation (of either narrow or broad bandwidth) or particle beam, of emitting electromagnetic radiation in a narrow wavelength band and/or absorbing energy and/or scattering or diffracting said excitation, thus permitting the simultaneous usage of a number of such probes emitting different wavelengths of electromagnetic radiation to thereby permit simultaneous detection of the presence of a number of detectable substances in a given material. The probe material is stable in the presence of light or oxygen, capable of being excited by energy over a wide spectrum, and has a narrow band of emission, resulting in an improved material and process for the simultaneous and/or sequential detection of a number of detectable substances in a material such as a biological material.

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Having thus described the invention what is claimed is:

1. A semiconductor nanocrystal compound, comprising:
  - a) a water-soluble semiconductor nanocrystal comprising:
    - i) a core comprising a first semiconductor material; and
    - ii) a core-overcoating shell comprising a second semiconductor material; and
  - b) a linking agent linked to said water-soluble semiconductor nanocrystal and capable of linking to an affinity molecule.
2. The compound of claim 1, wherein said first semiconductor material is a II–VI semiconductor or a III–V semiconductor.
3. The compound of claim 2, wherein said first semiconductor material is a II–VI semiconductor.
4. The compound of claim 2, wherein said first semiconductor material is a III–V semiconductor.
5. The compound of claim 3, wherein said first semiconductor material is MgS, MgSe, MgTe, CaS, CaSe, CaTe, SrS, SrSe, SrTe, BaS, BaSe, BaTe, ZnS, ZnSe, ZnTe, CdS, CdSe, CdTe, HgS, HgSe, or HgTe.
6. The compound of claim 4, wherein said first semiconductor material is GaAs, InGaAs, InP, or InAs.
7. The compound of claim 1, wherein said second semiconductor material is a II–VI semiconductor or a III–V semiconductor.
8. The compound of claim 7, wherein said second semiconductor material is a II–VI semiconductor.
9. The compound of claim 7, wherein said second semiconductor material is a III–V semiconductor.
10. The compound of claim 8, wherein said second semiconductor material is MgS, MgSe, MgTe, CaS, CaSe, CaTe, SrS, SrSe, SrTe, BaS, BaSe, BaTe, ZnS, ZnSe, ZnTe, CdS, CdSe, CdTe, HgS, HgSe, or HgTe.
11. The compound of claim 9, wherein said second semiconductor material is GaAs, InGaAs, InP, or InAs.
12. The compound of claim 1, wherein said first semiconductor material is CdSe and the second semiconductor material is ZnS.
13. The compound of claim 1, wherein said linking agent comprises a thiol moiety.
14. The compound of claim 13, wherein said linking agent further comprises an alkyl group.
15. The compound of claim 14, wherein said alkyl group is a propyl group.
16. The compound of claim 1, wherein said linking agent is N-(3-aminopropyl)-3-mercapto-benzamide, 3-aminopropyl-trimethoxysilane, 3-mercaptopropyl-trimethoxysilane, 3-maleimidopropyl-trimethoxysilane, or 3-hydrazidopropyl-trimethoxysilane.
17. The compound of claim 1, wherein said nanocrystal compound further comprises a glass coating on said shell.
18. The compound of claim 17, wherein said glass coating comprises a polymeric oxide.
19. The compound of claim 18, wherein said polymeric oxide is an oxide of silicon, an oxide of boron, an oxide of phosphorus, or a mixture thereof.
20. The compound of claim 18, wherein said glass coating further comprises a metal silicate, a metal borate or a metal phosphate.
21. The compound of claim 7, wherein said linking agent is N-(3-aminopropyl)-3-mercapto-benzamide, 3-aminopropyl-trimethoxysilane, 3-mercaptopropyl-trimethoxysilane, 3-maleimidopropyl-trimethoxysilane, or 3-hydrazidopropyl-trimethoxysilane.
22. The compound of claim 1, wherein said shell epitaxially surrounds said core.
23. A luminescent semiconductor nanocrystal compound, comprising:

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- a) a water-soluble luminescent semiconductor nanocrystal comprising:
  - i) a core comprising a first semiconductor material; and
  - ii) a core-overcoating shell comprising a second semiconductor material; and
- b) a linking agent linked to said water-soluble luminescent semiconductor nanocrystal and capable of linking to an affinity molecule.

24. A luminescent semiconductor nanocrystal compound, comprising:

- a) a water-soluble luminescent semiconductor nanocrystal comprising:
  - i) a core comprising a first luminescent semiconductor nanocrystal material; and
  - ii) a core-overcoating shell comprising a second semiconductor material; and
- b) a linking agent linked to said water-soluble luminescent semiconductor nanocrystal and capable of linking to an affinity molecule.

25. A luminescent semiconductor nanocrystal compound, comprising:

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- a) a water-soluble luminescent semiconductor nanocrystal comprising:
  - i) a core comprising a first semiconductor material; and
  - ii) a core-overcoating shell comprising a second luminescent semiconductor nanocrystal material; and
- b) a linking agent lined to said water-soluble luminescent semiconductor nanocrystal and capable of linking to an affinity molecule.

26. A luminescent semiconductor nanocrystal compound, comprising:

- a) a water-soluble luminescent semiconductor nanocrystal comprising:
  - i) a core comprising a first luminescent semiconductor nanocrystal material; and
  - ii) a core-overcoating shell comprising a second luminescent semiconductor nanocrystal material; and
- b) a linking agent linked to said water-soluble luminescent semiconductor nanocrystal and capable of linking to an affinity molecule.

\* \* \* \* \*

## **EXHIBIT 2**



US006699723B1

(12) **United States Patent**  
Weiss et al.

(10) **Patent No.:** US 6,699,723 B1  
(45) **Date of Patent:** Mar. 2, 2004

(54) **ORGANO LUMINESCENT SEMICONDUCTOR NANOCRYSTAL PROBES FOR BIOLOGICAL APPLICATIONS AND PROCESS FOR MAKING AND USING SUCH PROBES**

FOREIGN PATENT DOCUMENTS

EP	0 990 903	4/2000	.....	G01N/33/58
WO	WO 98/04740	2/1998	.....	C12Q/1/68
WO	WO 99/19515	4/1999	.....	C12Q/1/68

OTHER PUBLICATIONS

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Alivisatos, A. P., "Semiconductor Clusters, Nanocrystals, and Quantum Dots," *Science* 271 (Feb. 16, 1996):933-937.

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Alivisatos, A. P., "Perspectives on the Physical Chemistry of Semiconductor Nanocrystals," *J. Phys. Chem.* 100 (1996):13226-13239.

(\* ) Notice: Subject to any disclaimer, the term of this patent is extended or adjusted under 35 U.S.C. 154(b) by 352 days.

Alivisatos, A. Paul, et al., "Organization of 'Nanocrystal Molecules' Using DNA," *Nature* 382 (Aug. 15, 1996):609-611.

(List continued on next page.)

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**Related U.S. Application Data**

(62) Division of application No. 09/349,833, filed on Jul. 8, 1999, now Pat. No. 6,423,551, which is a continuation of application No. 08/978,450, filed on Nov. 25, 1997, now Pat. No. 5,990,479.

(57) **ABSTRACT**

A semiconductor nanocrystal compound is described capable of linking to an affinity molecule. The compound comprises (1) a semiconductor nanocrystal capable of emitting electromagnetic radiation and/or absorbing energy, and/or scattering or diffracting electromagnetic radiation—when excited by an electromagnetic radiation source or a particle beam; and (2) at least one linking agent, having a first portion linked to the semiconductor nanocrystal and a second portion capable of linking to an affinity molecule. The compound is linked to an affinity molecule to form a semiconductor nanocrystal probe capable of bonding with a detectable substance. Subsequent exposure to excitation energy will excite the semiconductor nanocrystal in the probe, causing the emission of electromagnetic radiation. Further described are processes for respectively: making the semiconductor nanocrystal compound; making the semiconductor nanocrystal probe; and using the probe to determine the presence of a detectable substance in a material.

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(52) **U.S. Cl.** ..... **436/518**; 424/9.32; 424/9.34; 424/9.341; 424/9.36; 428/402; 428/402.24; 428/403; 428/404; 428/405; 428/406; 436/172; 436/173; 436/524; 436/525; 436/527

(58) **Field of Search** ..... 424/9.32, 9.34, 424/9.341, 9.36; 428/402, 402.24, 403, 404, 405, 406; 436/172, 173, 518, 524, 525, 527

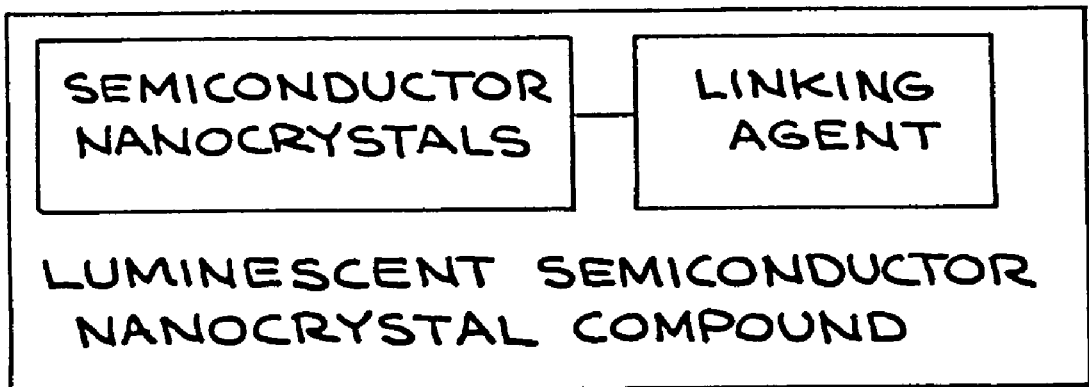
(56) **References Cited**

U.S. PATENT DOCUMENTS

3,996,345 A	12/1976	Ullman et al. ....	424/12
4,637,988 A	1/1987	Hinshaw et al. ....	436/546
4,777,128 A	10/1988	Lippa .....	435/5

(List continued on next page.)

**27 Claims, 3 Drawing Sheets**



## US 6,699,723 B1

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## U.S. PATENT DOCUMENTS

5,262,357 A	11/1993	Alivisatos et al. ....	437/233
5,319,209 A	6/1994	Miyakawa et al. ....	250/459.1
5,460,831 A	10/1995	Kossovsky et al. ....	424/493
5,505,928 A	4/1996	Alivisatos et al. ....	423/299
5,537,000 A	7/1996	Alivisatos et al. ....	313/506
5,585,640 A	12/1996	Huston et al. ....	250/483.1
5,674,698 A	10/1997	Zarling et al. ....	435/7.92
5,736,330 A	4/1998	Fulton .....	435/6
5,751,018 A	5/1998	Alivisatos et al. ....	257/64

## OTHER PUBLICATIONS

Mahtab, Rahina, et al., "Protein-Sized Quantum Dot Luminescence Can Distinguish Between 'Straight,' 'Bent,' and 'Kinked' Oligonucleotides," *J. Am. Chem. Soc.* 117 (1995):9099-9100.

Murphy, Catherine J., et al., "Quantum Dots as Inorganic DNA-Binding Proteins," *Mat. Res. Soc. Symp. Proc.* 452 (1997):597-600.

Peng, Xiaogang, et al., "Epitaxial Growth of Highly Luminescent CdSe/Cds Core/Shell Nanocrystals with Photostability and Electronic Accessibility," *Journal of the American Chemical Society*, vol. 119, No. 30, pp. 7019-7029.

Peng, Xiaogang, et al., "Synthesis and Isolation of a Homodimer of Cadmium Selenide Nanocrystals," *Angewandte Chemie-International Edition in English*, 36 (1997):145-147.

Service, Robert F., "Semiconductor Beacons Light Up Cell Structures," *Science* 281 (Sep. 25, 1998):1930-1931.

Shröck, E., et al., "Multicolor Spectral Karyotyping of Human Chromosomes," *Science* 273 (Jul. 26, 1996):494-497.

Zhang, Yu-zhong, et al., "Novel Flow Cytometry Compensation Standards: Internally Stained Fluorescent Microspheres with Matched Emission Spectra and Long-Term Stability," *Cytometry* 33 (1998):244-248.

Lacoste, T.D., et al., "Super Resolution Molecular Ruler Using Single Quantum Dots," *Biophysical Journal*, vol. 78, Jan., 2000, p. 402A, XP-000933548 Abstract.

Bruchez, Marcel, Jr., et al., "Semiconductor Nanocrystals as Fluorescent Biological Labels," *Science*, vol. 281, Sep. 25, 1998, pp. 2013-2016.

Beverloo, H.B., et al., "Preparation and Microscopic Visualization of Multicolor Luminescent Immunophosphors," Chapter 4 of Beverloo, H.B., "Inorganic Crystals as Luminescent Labels: Their Applications in Immunocytochemistry and Time-Resolved Microscopy," Ph.D. dissertation, University of Leiden (The Netherlands), May 13, 1992, pp. 553-573.

Bruchez, Marcel P., Jr., "Luminescent Semiconductor Nanocrystals: Intermittent Behavior and Use as Fluorescent Biological Probes," Ph.D. dissertation, University of California, Dec. 17, 1998.

Bruchez, Marcel, Jr., et al., "Semiconductor Nanocrystals as Fluorescent Probes for Biology," *Cytometry Supp.* 9 (1998):26.

Chan, Warren C.W., et al., "Quantum Dot Bioconjugates for Ultrasensitive Nonisotopic Detection," *Science* 281 (Sep. 25, 1998):2016-2018.

Coffer, Jeffrey L., et al., "Characterization of Quantum-Confined CdS Nanocrystallites Stabilized by Deoxyribonucleic Acid (DNA)," *Nanotechnol.* 3 (1992):69-76.

Cook, Neil D., "Scintillation Proximity Assay: A Versatile High-Throughput Screening Technology," *Drug Discovery Today* 1 (Jul., 1996):287-294.

Correa-Duarte, Miguel A., et al., "Stabilization of CdS Semiconductor Nanoparticles Against Photodegradation by a Silica Coating Procedure," *Chem. Phys. Lett.* 286 (Apr. 17, 1998):497-501.

Dabbousi, B.O., et al., "(CdSe)ZnS Core-Shell Quantum Dots: Synthesis and Characterization of a Size Series of Highly Luminescent Nanocrystal-lites," *Journal of Physical Chemistry B*, vol. 101, 1997, pp. 9463-9475.

Jacoby, Mitch, "Quantum Dots Meet Biomolecules," *C&E News* 76 (Sep. 28, 1998):Copied from the Internet as pp. 1-3.

Kagan, C.R., et al., "Electronic Energy Transfer in CdSe Quantum Dot Solids," *Phys. Rev. Lett.* 76 (Feb. 26, 1996):1517-1520.

Leff, David N., "Color-Coding Quantum Dots Debut with Promising Careers in Clinical Diagnostics Field," *Bioworld Today*, Sep. 25, 1998, Copied from the Internet as pp. 1-2.

Liz-Marzán, Luis M., et al., "Synthesis of Nanosized Gold-Silica Core-Shell Particles," *Lanqmuir* 12 (1996):4329-4335.

Mahtab, Rahina, et al., "Preferential Adsorption of a 'Kinked' DNA to a Neutral Curved Surface: Comparisons to and Implications for Nonspecific DNA-Protein Interactions," *J. Am. Chem. Soc.* 118 (1996):7028-7032.



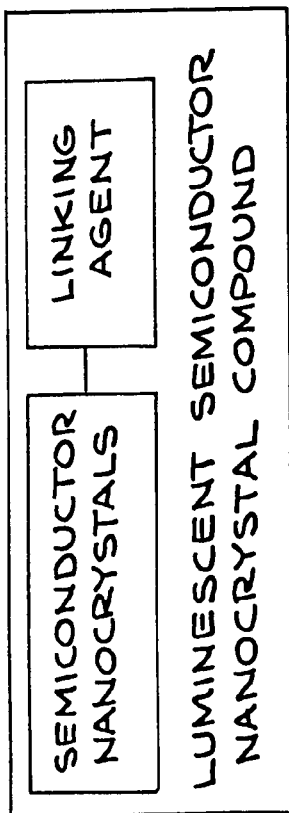


FIG. 1

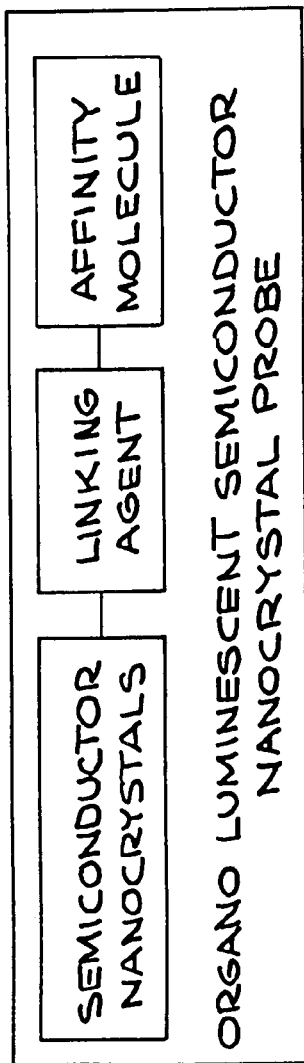


FIG. 2

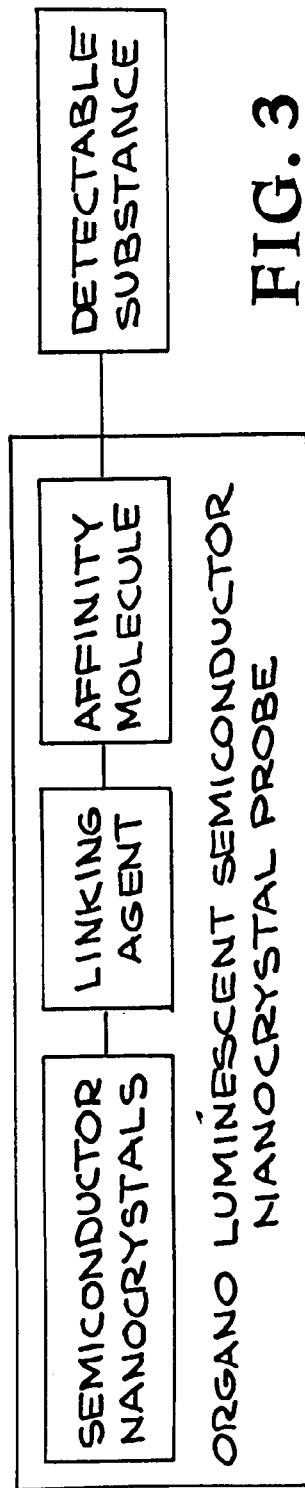


FIG. 3

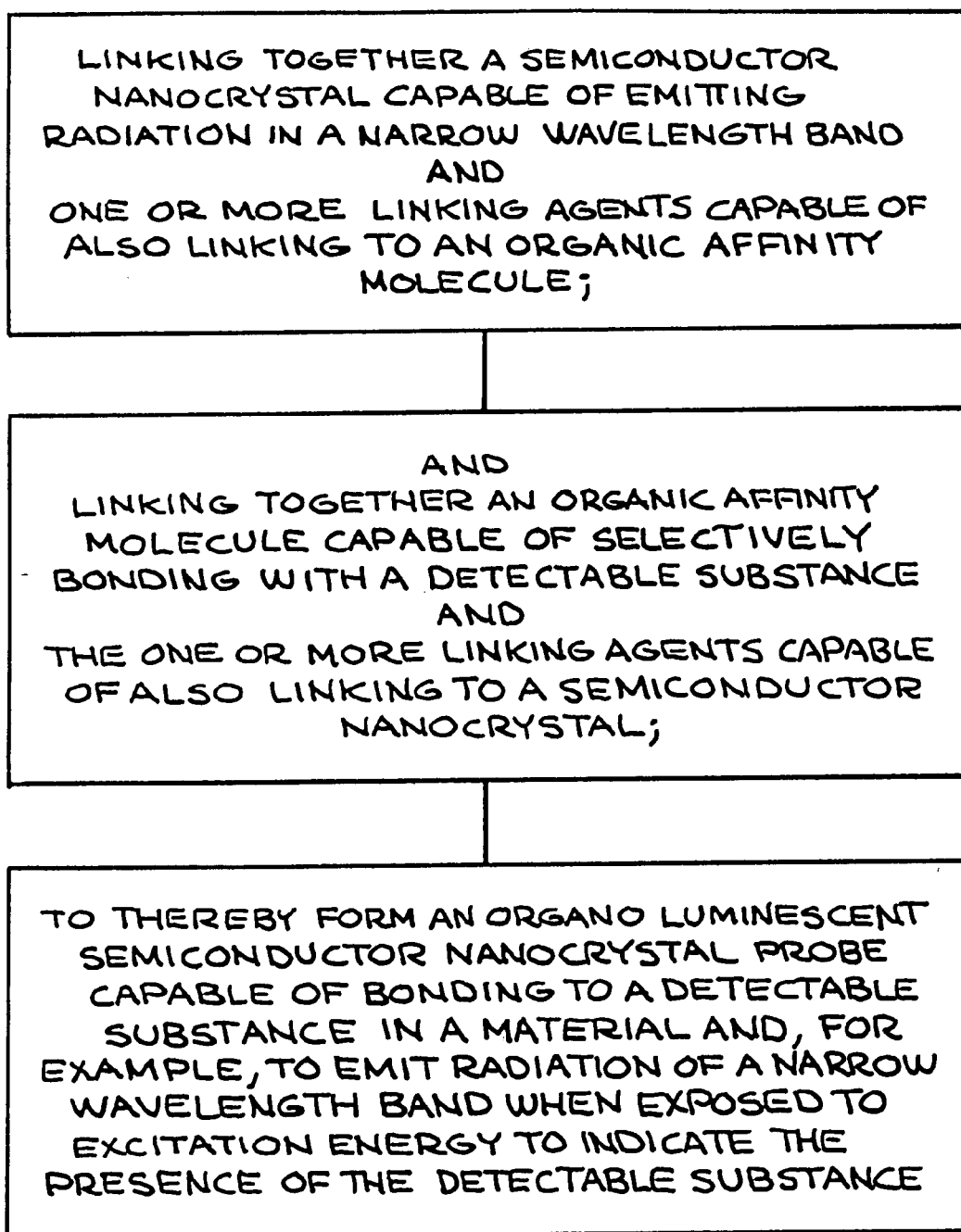


FIG. 4

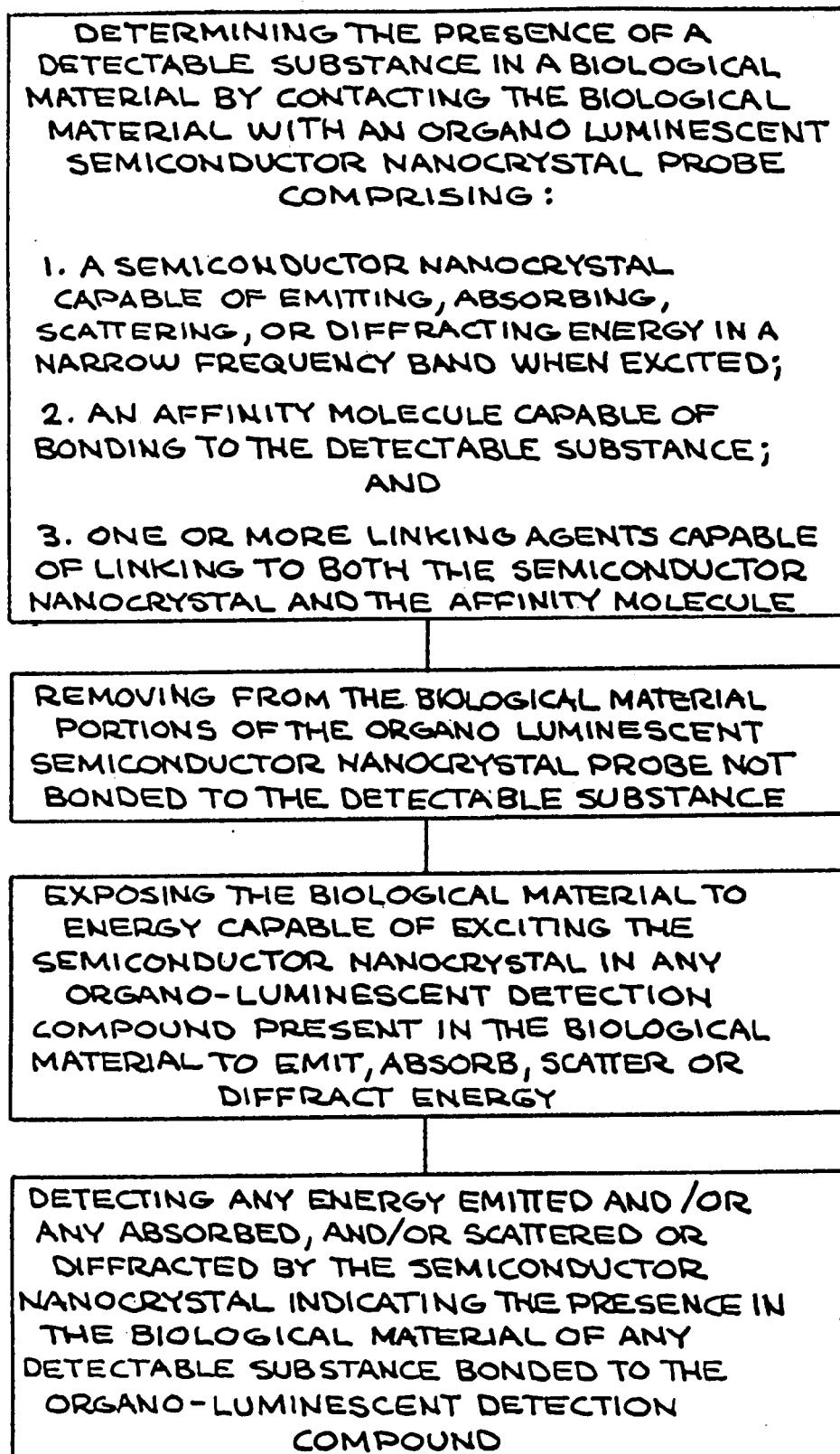


FIG. 5

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**ORGANO LUMINESCENT  
SEMICONDUCTOR NANOCRYSTAL PROBES  
FOR BIOLOGICAL APPLICATIONS AND  
PROCESS FOR MAKING AND USING SUCH  
PROBES**

CROSS-REFERENCE TO RELATED  
APPLICATION

“This application is a divisional application of U.S. patent application Ser. No. 09/349,833 filed Jul. 8, 1999, now U.S. Pat. No. 6,423,551 which is a continuation of U.S. patent application Ser. No. 08/978,450 filed Nov. 25, 1997, now U.S. Pat. No. 5,990,479 issued Nov. 23, 1999.”

The invention described herein arose in the course of, or under, Contract No. DE-AC03-SF00098 between the United States Department of Energy and the University of California for the operation of the Ernest Orlando Lawrence Berkeley National Laboratory. The Government may have rights to the invention.

BACKGROUND OF THE INVENTION

1. Field of the Invention

This invention relates to organo luminescent semiconductor nanocrystal probes for biological applications wherein the probes includes a plurality of semiconductor nanocrystals capable of luminescence and/or absorption and/or scattering or diffraction when excited by a radiation or particle beam.

2. Description of the Related Art

Fluorescent labeling of biological systems is a well known analytical tool used in modern bio-technology as well as analytical chemistry. Applications for such fluorescent labeling include technologies such as medical (and non-medical) fluorescence microscopy, histology, flow cytometry, fluorescence in-situ hybridization (medical assays and research), DNA sequencing, immuno-assays, binding assays, separation, etc.

Conventionally, such fluorescent labeling involves the use of an organic dye molecule bonded to a moiety which, in turn, selectively bonds to a particular biological system, the presence of which is then identified by excitation of the dye molecule to cause it to fluoresce. There are a number of problems with such an analytical system. In the first place, the emission of light of visible wavelengths from an excited dye molecule usually is characterized by the presence of a broad emission spectrum as well as a broad tail of emissions on the red side of the spectrum, i.e., the entire emission spectrum is rather broad. As a result, there is a severe limitation on the number of different color organic dye molecules which may be utilized simultaneously or sequentially in an analysis since it is difficult to either simultaneously or even non-simultaneously detect or discriminate between the presence of a number of different detectable substances due to the broad spectrum emissions and emission tails of the labelling molecules. Another problem is that most dye molecules have a relatively narrow absorption spectrum, thus requiring either multiple excitation beams used either in tandem or sequentially for multiple wavelength probes, or else a broad spectrum excitation source which is sequentially used with different filters for sequential excitation of a series of probes respectively excited at different wavelengths.

Another problem frequently encountered with existing dye molecule labels is that of photostability. Available fluorescent molecules bleach, or irreversibly cease to emit

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light, under repeated excitation ( $10^4$ – $10^8$ ) cycles of absorption/emission. These problems are often surmounted by minimizing the amount of time that the sample is exposed to light, and by removing oxygen and/or other radical species from the sample.

In addition, the probe tools used for the study of these systems by electron microscopy techniques are completely different from the probes used for study by fluorescence. Thus, it is not possible to label a material with a single type of probe for both electron microscopy and for fluorescence.

It would, therefore, be desirable to provide a stable probe material for biological applications having a wide absorption band and capable of exhibiting either a detectable change in absorption or of emitting radiation in a narrow wavelength band, without the presence of the large red emission tails characteristic of dye molecules (thereby permitting the simultaneous use of a number of such probe materials, each emitting light of a different narrow wavelength band) and/or capable of scattering or diffracting radiation. It would also be equally desirable to provide a single, stable probe material which can be used to image the same sample by both light and electron microscopy.

SUMMARY OF THE INVENTION

The invention comprises a luminescent semiconductor nanocrystal compound capable of linking to an affinity molecule to form an organo luminescent semiconductor nanocrystal probe capable of luminescence and/or absorption and/or scattering or diffracting when excited by an electromagnetic radiation source (of broad or narrow bandwidth) or a particle beam, and capable of exhibiting a detectable change in absorption and/or of emitting radiation in a narrow wavelength band and/or scattering or diffracting when so excited. The luminescent semiconductor nanocrystal compound preferably comprises: (1) a semiconductor nanocrystal capable of luminescence and/or absorption and/or scattering or diffraction when excited by an electromagnetic radiation source (of broad or narrow bandwidth) or a particle beam, and capable of exhibiting a detectable change in absorption and/or of emitting radiation in a narrow wavelength band and/or scattering or diffracting when excited; and (2) a linking agent having a first portion linked to the semiconductor nanocrystal, and a second portion capable of linking to an affinity molecule.

The invention further comprises an organo luminescent semiconductor nanocrystal probe formed by linking the above described luminescent semiconductor nanocrystal compound to an affinity molecule capable of bonding to a detectable substance in a material. As a result the organo luminescent semiconductor nanocrystal probe, in one embodiment, is capable of absorbing or scattering or diffracting energy from either a particle beam or an electromagnetic radiation source (of broad or narrow bandwidth), and is capable of emitting electromagnetic radiation in a narrow wavelength band when so excited; while in another embodiment the amount of energy so absorbed, or scattered, or diffracted from either a particle beam or an electromagnetic radiation source (of broad or narrow bandwidth), is detectable, i.e., the change in absorption, scattering, or diffraction is detectable.

Therefore, treatment of a material with the organo luminescent semiconductor nanocrystal probe, and subsequent exposure of this treated material to excitation energy (from either a particle beam or an electromagnetic radiation source of broad or narrow bandwidth) to determine the presence of the detectable substance within the material, will excite the

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semiconductor nanocrystals in the organo luminescent semiconductor nanocrystal probe bonded to the detectable substance, resulting in the emission of electromagnetic radiation of a narrow wavelength band and/or a detectable change in the amount of energy being absorbed and/or scattered or diffracted, signifying the presence, in the material, of the detectable substance bonded to the organo luminescent semiconductor nanocrystal probe.

The invention also comprises a process for making the luminescent semiconductor nanocrystal compound and for making the organo luminescent semiconductor nanocrystal probe comprising the luminescent semiconductor nanocrystal compound linked to an affinity molecule capable of bonding to a detectable substance. The organo luminescent semiconductor nanocrystal probe of the invention is stable with respect to repeated excitation by light, or exposure to oxygen or other radicals. The invention further comprises a process for treating a material, such as a biological material, to determine the presence of a detectable substance in the material which comprises contacting the material with the organo luminescent semiconductor nanocrystal probe, removing from the material portions of the organo luminescent semiconductor nanocrystal probe not bonded to the detectable substance, and then exposing the material to activation energy from either an electromagnetic radiation source (of broad or narrow bandwidth) or a particle beam. The presence of the detectable substance in the material is then determined either by measuring the absorption of energy by the organo luminescent semiconductor nanocrystal probe and/or detecting the emission of radiation of a narrow wavelength band by the organo luminescent semiconductor nanocrystal probe and/or detecting the scattering or diffraction by the organo luminescent semiconductor nanocrystal probe, indicative (in either case) of the presence of the organo luminescent semiconductor nanocrystal probe bonded to the detectable substance in the material.

### BRIEF DESCRIPTION OF THE DRAWINGS

FIG. 1 is a block diagram of the luminescent semiconductor nanocrystal compound of the invention.

FIG. 2 is a block diagram of the organo luminescent semiconductor nanocrystal probe of the invention.

FIG. 3 is a block diagram showing the affinity between a detectable substance and the organo luminescent semiconductor nanocrystal probe of the invention.

FIG. 4 is a flow sheet illustrating the process of forming the organo luminescent semiconductor nanocrystal probe of the invention.

FIG. 5 is a flow sheet illustrating a typical use of the organo luminescent semiconductor nanocrystal probe of the invention in detecting the presence of a detectable substance in a material such as a biological, material.

### DETAILED DESCRIPTION OF THE INVENTION

In The invention comprises a luminescent semiconductor nanocrystal compound capable of linking to an organic molecule and capable of exhibiting a detectable change in absorption and/or of emitting electromagnetic radiation in a narrow wavelength band and/or scattering or diffracting when excited by either an electromagnetic radiation source (of broad or narrow bandwidth) or a particle beam. The luminescent semiconductor nanocrystal compound, in turn, comprises: (1) semiconductor nanocrystals capable of exhibiting a detectable change in absorption and/or of emitting

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electromagnetic radiation in a narrow wavelength band when excited by either an electromagnetic radiation source (of broad or narrow bandwidth) or a particle beam; and (2) one or more linking agents each having a first portion linked to the semiconductor nanocrystal and a second portion capable of linking to an organic affinity molecule.

The invention also comprises the above described luminescent semiconductor nanocrystal compound linked to the organic affinity molecule (through the linking agent) to form an organo luminescent semiconductor nanocrystal probe capable of bonding to a detectable substance and capable of exhibiting a detectable change in absorption and/or of emitting electromagnetic radiation in a narrow wavelength band and/or scattering or diffracting when excited by either an electromagnetic radiation source (of broad or narrow bandwidth) or a particle beam. Treatment of a material (typically a biological material) with the organo luminescent semiconductor nanocrystal probe, and subsequent exposure of this treated material to excitation energy, as described above, to determine the presence of the detectable substance within the material, will excite the semiconductor nanocrystal in the organo luminescent semiconductor nanocrystal probe bonded to the detectable substance, causing the detectable change in absorption and/or emission of electromagnetic radiation of a narrow wavelength band and/or scattering or diffraction signifying (in either instance) the presence in the material, of the detectable substance bonded to the organo luminescent semiconductor nanocrystal probe.

The invention also comprises a process for making the luminescent semiconductor nanocrystal compound, and a process for making the organo luminescent semiconductor nanocrystal probe comprising the luminescent semiconductor nanocrystal compound linked to an affinity molecule capable of bonding to a detectable substance.

The invention further comprises a process for treating a material, such as a biological material, to determine the presence of a detectable substance in the material which comprises: (1) contacting the material with the organo luminescent semiconductor nanocrystal probe, (2) removing from the material portions of the organo luminescent semiconductor nanocrystal probe not bonded to the detectable substance, (3) exposing the material to energy (such as the above-described electromagnetic energy source or particle beam) capable of exciting the semiconductor nanocrystal to cause a detectable change in absorption and/or emission of electromagnetic radiation of a narrow wavelength band and/or scattering or diffraction signifying (in either instance) the presence of the organo luminescent semiconductor nanocrystal probe bonded to the detectable substance in the material, and (4) detecting either the change in absorbed energy or the electromagnetic radiation emitted or the scattering or diffraction by the semiconductor nanocrystal in the organo luminescent semiconductor nanocrystal probe.

#### a. Definitions

By use of the terms "nanometer crystal" or "nanocrystal" herein is meant an organic or inorganic single crystal particle having an average cross-section no larger than about 20 nanometers (nm) or  $20 \times 10^{-9}$  meters (200 Angstroms), preferably no larger than about 10 nm (100 Angstroms) and a minimum average cross-section of about 1 nm, although in some instances a smaller average cross-section nanocrystal, i.e., down to about 0.5 nm (5 Angstroms), may be acceptable. Typically the nanocrystal will have an average cross-section ranging in size from about 1 nm (10 Angstroms) to about 10 nm (100 angstroms).

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By use of the term “semiconductor nanocrystal” is meant a nanometer crystal or nanocrystal of Group II–VI and Group III–V semiconductor compounds capable of emitting electromagnetic radiation upon excitation, although the use of Group IV semiconductors such as germanium or silicon, or the use of organic semiconductors, may be feasible under certain conditions.

By use of the term “a narrow wavelength band”, with regard to the electromagnetic radiation emission of the semiconductor nanocrystal, is meant a wavelength band of emissions not exceeding about 40 nm, and preferably not exceeding about 20 nm in width and symmetric about the center, in contrast to the emission bandwidth of about 100 nm for a typical dye molecule, with a red tail which may extend the band width out as much as another 100 nm. It should be noted that the bandwidths referred to are determined from measurement of the width of the emissions at half peak height (FWHM), and are appropriate in the range of 200 nm to 2000 nm.

By use of the term “a broad absorption band”, with regard to the electromagnetic radiation absorption of the semiconductor nanocrystal is meant a continuously increasing absorption from the onset, which occurs near to, but at slightly higher energy than the “narrow wavelength band” of the emission. This is in contrast to the “narrow absorption band” of dye molecules which occurs near the emission peak on the high energy side, but drops off rapidly away from that wavelength.

By use of the term “detectable substance” is meant an entity or group, the presence or absence of which in a material such as a biological material, is to be ascertained by use of the organo-luminescent semiconductor nanocrystal probe of the invention.

By use of the term “affinity molecule” is meant the portion of the organo luminescent semiconductor nanocrystal probe of the invention which will selectively bond to a detectable substance (if present) in the material (e.g., biological material) being analyzed.

By use of the term “linking agent” is meant a substance capable of linking with a semiconductor nanocrystal and also capable of linking to an affinity molecule.

The terms “link” and “linking” are meant to describe the adherence between the affinity molecule and the semiconductor nanocrystals, either directly or through a moiety identified herein as a linking agent. The adherence may comprise any sort of bond, including, but not limited to, covalent, ionic, hydrogen bonding, Van der Waals’ forces, or mechanical bonding, etc.

The terms “bond” and “bonding” are meant to describe the adherence between the affinity molecule and the detectable substance. The adherence may comprise any sort of bond, including, but not limited to, covalent, ionic, or hydrogen bonding, Van der Waals’ forces, or mechanical bonding, etc.

The term “luminescent semiconductor nanocrystal compound”, as used herein, is intended to define a semiconductor nanocrystal linked to one or more linking agents and capable of linking to an affinity molecule, while the term “organo-luminescent semiconductor nanocrystal probe” is intended to define a luminescent semiconductor nanocrystal compound linked to an affinity molecule.

The term “glass” as used herein is intended to include one or more oxides of silicon, boron, and/or phosphorus, or a mixture thereof, as well as the further optional inclusion of one or more metal silicates, metal borates or metal phosphates therein.

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#### b. The Semiconductor Nanocrystals

The semiconductor nanocrystals useful in the practice of the invention include nanocrystals of Group II–VI semiconductors such as MgS, MgSe, MgTe, CaS, CaSe, CaTe, SrS, SrSe, SrTe, BaS, BaSe, BaTe, ZnS, ZnSe, ZnTe, CdS, CdSe, CdTe, HgS, HgSe, and HgTe; and nanocrystals of Group III–V semiconductors such as GaAs, InGaAs, InP, and InAs. As mentioned above, the use of Group IV semiconductors such as germanium or silicon, or the use of organic semiconductors, may also be feasible under certain conditions.

Formation of nanometer crystals of Group III–V semiconductors is described in copending and commonly assigned Alivisatos et al. U.S. Pat. No. 5,751,018; Alivisatos et al. U.S. Pat. No. 5,505,928; and Alivisatos et al. U.S. Pat. No. 5,262,357, which also describes the formation of Group II–VI semiconductor nanocrystals, and which is also assigned to the assignee of this invention. Also described therein is the control of the size of the semiconductor nanocrystals during formation using crystal growth terminators. The teachings of Alivisatos et al. U.S. Pat. No. 5,751,018 and Alivisatos et al. U.S. Pat. No. 5,262,357 are each hereby specifically incorporated by reference.

In a preferred embodiment, the nanocrystals are used in a core/shell configuration wherein a first semiconductor nanocrystal forms a core ranging in diameter, for example, from about 20 Å to about 100 Å, with a shell of another semiconductor nanocrystal material grown over the core nanocrystal to a thickness of, for example, 1–10 monolayers in thickness. When, for example, a 1–10 monolayer thick shell of CdS is epitaxially grown over a core of CdSe, there is a dramatic increase in the room temperature photoluminescence quantum yield. Formation of such core/shell nanocrystals is described more fully in a publication by one of us with others entitled “Epitaxial Growth of Highly Luminescent CdSe/CdS Core/Shell Nanocrystals with Photostability and Electronic Accessibility”, by Peng, Schiamp, Kadavanich, and Alivisatos, published in the Journal of the American Chemical Society, Volume 119, No. 30. 1997, at pages 7019–7029, the subject matter of which is hereby specifically incorporated herein by reference.

The semiconductor nanocrystals used in the invention will have a capability of emitting light within a narrow wavelength band of about 40 nm or less, preferably about 20 nm or less, thus permitting the simultaneous use of a plurality of differently colored organo luminescent semiconductor nanocrystal probes with different semiconductor nanocrystals without overlap (or with a small amount of overlap) in wavelengths of emitted light (unlike the use of dye molecules with broad emission lines (e.g., ~100 nm) and broad tails of emission (e.g., another 100 nm) on the red side of the spectrum), thus allowing for the simultaneous detection of a plurality of detectable substances.

#### c. Affinity Molecule

The particular affinity molecule forming a part of the organo-luminescent semiconductor nanocrystal probe of the invention will be selected based on its affinity for the particular detectable substance whose presence or absence, for example, in a biological material, is to be ascertained. Basically, the affinity molecule may comprise any molecule capable of being linked to a luminescent semiconductor nanocrystal compound which is also capable of specific recognition of a particular detectable substance. In general, any affinity molecule useful in the prior art in combination with a dye molecule to provide specific recognition of a

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detectable substance will find utility in the formation of the organo-luminescent semiconductor nanocrystal probes of the invention. Such affinity molecules include, by way of example only, such classes of substances as monoclonal and polyclonal antibodies, nucleic acids (both a monomeric and oligomeric), proteins, polysaccharides, and small molecules such as sugars, peptides, drugs, and ligands. Lists of such affinity molecules are available in the published literature such as, by way of example, the "Handbook of Fluorescent Probes and Research Chemicals", (sixth edition) by R. P. Haugland, available from Molecular Probes, Inc.

#### d. The Linking Agent

The organo-luminescent semiconductor nanocrystal probe of the invention will usually find utility with respect to the detection of one or more detectable substances in organic materials, and in particular to the detection of one or more detectable substances in biological materials. This requires the presence, in the organo-luminescent semiconductor nanocrystal probe, of an affinity molecule or moiety, as described above, which will bond the organo-luminescent semiconductor nanocrystal probe to the detectable substance in the organic/biological material so that the presence of the detectable material may be subsequently ascertained. However, since the semiconductor nanocrystals are inorganic, they may not bond directly to the organic affinity molecule. In these case therefore, there must be some type of linking agent present in the organo-luminescent semiconductor nanocrystal probe which is capable of forming a link to the inorganic semiconductor nanocrystal as well as to the organic affinity molecule in the organo-luminescent semiconductor nanocrystal probe.

One form in which the semiconductor nanocrystal may be linked to an affinity molecule via a linking agent is by coating the semiconductor nanocrystal with a thin layer of glass, such as silica ( $\text{SiO}_x$  where  $x=1-2$ ), using a linking agent such as a substituted silane, e.g., 3-mercaptopropyl-trimethoxy silane to link the nanocrystal to the glass. The glass-coated semiconductor nanocrystal may then be further treated with a linking agent, e.g., an amine such as 3-aminopropyl-trimethoxysilane, which will function to link the glass-coated semiconductor nanocrystal to the affinity molecule. That is, the glass-coated semiconductor nanocrystal may then be linked to the affinity molecule. It is within the contemplation of this invention that the original luminescent semiconductor nanocrystal compound may also be chemically modified after it has been made in order to link effectively to the affinity molecule. A variety of references summarize the standard classes of chemistry which may be used to this end, in particular the "Handbook of Fluorescent Probes and Research Chemicals", (6th edition) by R. P. Haugland, available from Molecular Probes, Inc., and the book "Bioconjugate Techniques", by Greg Hermanson, available from Academic Press, New York.

When the semiconductor nanocrystal is coated with a thin layer of glass, the glass, by way of example, may comprise a silica glass ( $\text{SiO}_x$  where  $x=1-2$ ), having a thickness ranging from about 0.5 nm to about 10 nm, and preferably from about 0.5 nm to about 2 nm.

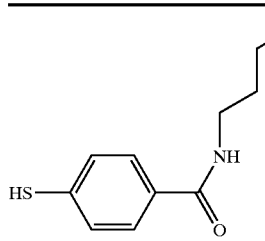
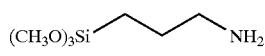
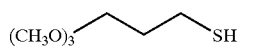
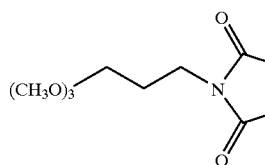
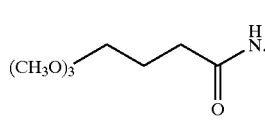
The semiconductor nanocrystal is coated with the coating of thin glass, such as silica, by first coating the nanocrystals with a surfactant such as tris-octyl-phosphine oxide, and then dissolving the surfactant-coated nanocrystals in a basic methanol solution of a linking agent, such as 3-mercaptopropyl-tri-methoxy silane, followed by partial hydrolysis which is followed by addition of a glass-affinity

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molecule linking agent such as amino-propyl trimethoxysilane which will link to the glass and serve to form a link with the affinity molecule.

When the linking agent does not involve the use of a glass coating on the semiconductor nanocrystal, it may comprise a number of different materials, depending upon the particular affinity molecule, which, in turn, depends upon the type of detectable material being analyzed for. It should also be noted that while an individual linking agent may be used to link to an individual semiconductor nanocrystal, it is also within the contemplation of the invention that more than one linking agent may bond to the same semiconductor nanocrystal and vice versa.

A few examples of the types of linking agents which may be used to link to both the semiconductor nanocrystal (or to a glass coating on the nanocrystal) and to the organic affinity molecule in the probe are illustrated in the table below, it being understood that this is not intended to be an exhaustive list:

Linking Agent	
Structure	Name
	N-(3-aminopropyl)3-mercaptopropylbenzamide
	3-aminopropyl-trimethoxysilane
	3-mercaptopropyl-trimethoxysilane
	3-maleimidopropyl-trimethoxysilane
	3-hydrazidopropyl-trimethoxysilane

It should be further noted that a plurality of polymerizable linking agents may be used together to form an encapsulating net or linkage around an individual nanocrystal (or group of nanocrystals). This is of particular interest where the particular linking agent is incapable of forming a strong bond with the nanocrystal. Examples of linking agents capable of bonding together in such a manner to surround the nanocrystal with a network of linking agents include, but are not limited to: diacetylenes, acrylates, acrylamides, vinyl, styryl, and the aforementioned silicon oxide, boron oxide, phosphorus oxide, silicates, borates and phosphates.

#### e. The Excitation of the Probe and Detection of Emission/Absorption

As previously mentioned, the organo luminescent semiconductor nanocrystal probe of the invention is capable of

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being excited over a broad bandwidth, yet exhibits emission in a narrow wavelength band, in contrast to the dye molecules used in the prior art. Thus electromagnetic radiation of wavelength ranging from x-ray to ultraviolet to visible to infrared waves may be used to excite the luminescent semiconductor nanocrystals in the probe. In addition, the luminescent semiconductor nanocrystals are capable of excitation from bombardment with a particle beam such as an electron beam (e-beam). Furthermore, because of the broad bandwidth at which the luminescent semiconductor nanocrystals are excitable, one may use a common excitation source for the simultaneous excitation of several probes, i.e., several probes which give off radiation at different frequencies, thus permitting simultaneous excitation and detection of the presence of several probes indicating, for example, the presence of several detectable substances in the material being examined.

Thus, for example, a laser radiation source of a given frequency, e.g., blue light, may be used to excite a first organo luminescent semiconductor nanocrystal probe capable of emitting radiation of a second frequency, e.g., red light, indicating the presence, in the material being illuminated, of a first detectable substance to which the particular red light-emitting organo luminescent semiconductor nanocrystal probe has bonded. At the same time, the same blue light laser source may also be exciting a second organo luminescent semiconductor nanocrystal probe (in the same material) capable of emitting radiation of a third frequency, e.g., green light, indicating the presence, in the material being illuminated, of a second detectable substance to which the particular green light-emitting organo luminescent semiconductor nanocrystal probe has bonded. Thus, unlike the prior art, multiple excitation sources need not be used (because of the broad bandwidth in which the organo luminescent semiconductor nanocrystal probe of the invention is capable of being excited), and the narrow band of emission of the specific semiconductor nanocrystals in each probe makes possible the elimination of sequencing and/or elaborate filtering to detect the emitted radiation.

With respect to the absorption of energy by the probe of the invention, when the excitation source is an electron beam, or an X-ray source, the presence of the organo luminescent semiconductor nanocrystal probe bonded to the detectable substance of interest in the material being analyzed can be ascertained using a commercially available energy absorption or scattering or diffraction detection system wherein changes in absorption or scattering cross section or in diffraction of the material being analyzed can be detected, signifying the presence of the probe in the material, which, in turn, indicates the presence of the detectable substance to which the probe is bonded in the material being analyzed. In addition, it may be possible to use electron or X-ray sources to detect the presence of the organo luminescent semiconductor nanocrystal probe bonded to the detectable substance by using a conventional detection system for the emission of visible light to observe the visible emission in the narrow wavelength of emission of the probe.

The following examples will serve to further illustrate the formation of the organo luminescent semiconductor nanocrystal probes of the invention, as well as their use in detecting the presence of a detectable substance in a material such as a biological material.

#### EXAMPLE 1

To illustrate the formation of the luminescent semiconductor nanocrystal compound (comprising the semiconduc-

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tor nanocrystals linked to a linking agent) 20 ml. of a 5 mM solution of (4-mercapto)benzoic acid was prepared with a pH of 10 using  $(\text{CH}_3)_4\text{NOH}\cdot 5\text{H}_2\text{O}$ . 20 mg of tris-octylphosphine oxide coated CdSe/CdS core/shell nanocrystals were added to the solution and stirred until completely dissolved. The resultant nanocrystal/linking agent solution was heated for 5 hours at 50–60° C. and then concentrated to a few ml by evaporation. Then an equal volume of acetone was added and the nanocrystals precipitate out of solution homogeneously. The precipitate was then washed with acetone, dried, and then can be stored.

The luminescent semiconductor nanocrystal compound prepared above can be linked with an appropriate affinity molecule to form the organo luminescent semiconductor nanocrystal probe of the invention to treat a biological material to determine the presence or absence of a detectable substance. That is, the luminescent semiconductor nanocrystal compound prepared above can be linked, for example, with avidin or streptavidin (as the affinity molecule) to form an organo luminescent semiconductor nanocrystal probe to treat a biological material to ascertain the presence of biotin; or the luminescent semiconductor nanocrystal compound prepared above can be linked with anti-digoxigenin to form an organo luminescent semiconductor nanocrystal probe to treat a biological material to ascertain the presence of digoxigenin.

#### EXAMPLE 2

To illustrate the formation of luminescent semiconductor nanocrystal compound (comprising glass-coated semiconductor nanocrystals linked to a linking agent), 50  $\mu\text{l}$  of 3-mercaptopropyl-trimethoxy silane was added to 40 ml of an anhydrous solution of 25 vol. % dimethylsulfoxide in methanol, and the pH was adjusted to 10–11 using  $(\text{CH}_3)_4\text{NOH}\cdot 5\text{H}_2\text{O}$ . 10 mg of tris-octylphosphine oxide coated CdSe/CdS core-shell particles, prepared by the technique described in the aforementioned Peng, Schlamp, Kadavanich, and Alivisatos article, were then dissolved in this solution, and stirred for several hours. The solution was diluted with 40 ml of methanol adjusted to a pH of 10 with  $(\text{CH}_3)_4\text{NOH}\cdot 5\text{H}_2\text{O}$ , and heated for 1 hour at 69° C. The solution was stirred for an hour, and 40 ml of a 90 vol. % methanol/9.89 vol. %  $\text{H}_2\text{O}$ /0.1 vol. % trimethoxysilylpropyl urea/0.01 vol. % aminopropyl-trimethoxy silane solution which had been stirring for at least an hour, was added, and stirred for 2 hours. Subsequently the reaction was heated to 69° C. for 15 minutes, and then cooled. 10 ml of a 10 vol. % chlorotrimethyl silane solution in methanol which had been adjusted to a pH of 10 using  $(\text{CH}_3)_4\text{NOH}\cdot 5\text{H}_2\text{O}$  was added, stirred for 2 hours, then heated to 60° C., and then partially concentrated under vacuum. Once the methanol had all evaporated, the solution was precipitated with acetone as an oil product comprising the luminescent semiconductor nanocrystal compound. The luminescent semiconductor nanocrystal compound may then be redissolved in water, and in a variety of buffer solutions to prepare it for linking it to an affinity molecule to form the organo luminescent semiconductor nanocrystal probe of the invention to treat a biological material to determine the presence or absence of a detectable substance.

Thus, the invention provides an organo luminescent semiconductor nanocrystal probe containing a semiconductor nanocrystal capable, upon excitation by either electromagnetic radiation (of either narrow or broad bandwidth) or particle beam, of emitting electromagnetic radiation in a narrow wavelength band and/or absorbing energy and/or scattering or diffracting said excitation, thus permitting the



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simultaneous usage of a number of such probes emitting different wavelengths of electromagnetic radiation to thereby permit simultaneous detection of the presence of a number of detectable substances in a given material. The probe material is stable in the presence of light or oxygen, capable of being excited by energy over a wide spectrum, and has a narrow band of emission, resulting in an improved material and process for the simultaneous and/or sequential detection of a number of detectable substances in a material such as a biological material.

Having thus described the invention what is claimed is:

1. A luminescent semiconductor nanocrystal probe, comprising:

- a) a water-soluble semiconductor-nanocrystal comprising:
  - i) a core comprising a first semiconductor material; and
  - ii) a core-overcoating shell comprising a second semiconductor material;
- b) a linking agent comprising a first portion and a second portion, wherein said first portion is linked to said water-soluble semiconductor nanocrystal; and
- c) an affinity molecule linked to said second portion of said linking agent.

2. The probe of claim 1, wherein said first semiconductor material is a II-VI semiconductor or a III-V semiconductor.

3. The probe of claim 2, wherein said first semiconductor material is a II-VI semiconductor.

4. The probe of claim 2, wherein said first semiconductor material is a III-V semiconductor.

5. The probe of claim 1, wherein said first semiconductor material is MgS, MgSe, MgTe, CaS, CaSe, CaTe, SrS, SrSe, SrTe, BaS, BaSe, BaTe, ZnS, ZnSe, ZnTe, CdS, CdSe, CdTe, HgS, HgSe, or HgTe.

6. The probe of claim 4, wherein said first semiconductor material is GaAs, InGaAs, InP, or InAs.

7. The probe of claim 1, wherein said second semiconductor material is a II-VI semiconductor or a III-V semiconductor.

8. The probe of claim 7, wherein said second semiconductor material is a II-VI semiconductor.

9. The probe of claim 7, wherein said second semiconductor material is a III-V semiconductor.

10. The probe of claim 8, wherein said second semiconductor material is MgS, MgSe, MgTe, CaS, CaSe, CaTe, SrS, SrSe, SrTe, BaS, BaSe, BaTe, ZnS, ZnSe, ZnTe, CdS, CdSe, CdTe, HgS, HgSe, or HgTe.

11. The probe of claim 9, wherein said second semiconductor material is GaAs, InGaAs, InP, or InAs.

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12. The probe of claim 1, wherein said first semiconductor material is CdSe and said second semiconductor material is ZnS.

13. The probe of claim 1, wherein said linking agent comprises a thiol moiety.

14. The probe of claim 13, wherein said linking agent further comprises an alkyl group.

15. The probe of claim 14, wherein said alkyl group is a propyl group.

16. The probe of claim 1, wherein said linking agent is chosen from N-(3-aminopropyl)-3-mercapto-benzamide, 3-aminopropyl-trimethoxysilane, 3-mercaptopropyl-trimethoxysilane, 3-maleimidopropyl-trimethoxysilane, and 3-hydrazidopropyl-trimethoxysilane.

17. The probe of claim 1, wherein said nanocrystal compound further comprises a glass coating on said shell.

18. The probe of claim 17, wherein said glass coating comprises a polymeric oxide.

19. The probe of claim 18, wherein said polymeric oxide is chosen from an oxide of silicon, an oxide of boron, an oxide of phosphorus, and a mixture thereof.

20. The probe of claim 18, wherein said glass coating further comprises metal silicate, a metal borate or a metal phosphate.

21. The probe of claim 17, wherein said linking agent is chosen from N-(3-aminopropyl)-3-mercapto-benzamide, 3-aminopropyl-trimethoxysilane, 3-mercaptopropyl-trimethoxysilane, 3-maleimidopropyl-trimethoxysilane, and 3-hydrazidopropyl-trimethoxysilane.

22. The probe of claim 1, wherein said affinity molecule is chosen from an antibody, a nucleic acid, a protein, a polysaccharide and a small molecule.

23. The probe of claim 1, wherein said affinity molecule is chosen from avidin, streptavidin, biotin and anti-digoxigenin.

24. The probe of claim 1, wherein said affinity molecule is streptavidin.

25. The probe of claim 1, wherein said linking agent is 3-mercaptopropyl-trimethoxysilane and said affinity molecule is chosen from avidin, streptavidin, biotin and anti-digoxigenin.

26. The probe of claim 1, wherein said linking agent is 3-mercaptopropyl-trimethoxysilane and said affinity molecule is streptavidin.

27. The probe of claim 1, wherein said shell epitaxially surrounds said core.

\* \* \* \* \*

## **EXHIBIT 3**



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(12) **United States Patent**  
Weiss et al.

(10) **Patent No.:** US **6,927,069 B2**  
(45) **Date of Patent:** \***Aug. 9, 2005**

(54) **ORGANO LUMINESCENT SEMICONDUCTOR NANOCRYSTAL PROBES FOR BIOLOGICAL APPLICATIONS AND PROCESS FOR MAKING AND USING SUCH PROBES**

(75) Inventors: **Shimon Weiss**, Pinole, CA (US); **Marcel Bruchez, Jr.**, Newark, CA (US); **Paul Alivisatos**, Oakland, CA (US)

(73) Assignee: **The Regents of the University of California**, Oakland, CA (US)

(\* ) Notice: Subject to any disclaimer, the term of this patent is extended or adjusted under 35 U.S.C. 154(b) by 298 days.

This patent is subject to a terminal disclaimer.

(21) Appl. No.: **10/155,759**

(22) Filed: **May 24, 2002**

(65) **Prior Publication Data**

US 2003/0100130 A1 May 29, 2003

**Related U.S. Application Data**

(63) Continuation of application No. 09/349,833, filed on Jul. 8, 1999, now Pat. No. 6,423,551, which is a continuation of application No. 08/978,450, filed on Sep. 25, 1997, now Pat. No. 5,990,479.

(51) **Int. Cl.**<sup>7</sup> ..... **G01N 33/543**

(52) **U.S. Cl.** ..... **436/518; 428/402; 428/402.24; 428/403; 428/404; 428/405; 428/406; 436/172; 436/173; 436/524; 436/525; 436/527**

(58) **Field of Search** ..... **428/402, 402.24, 428/403-406; 436/172, 173, 518, 524, 525, 527**

(56) **References Cited**

**U.S. PATENT DOCUMENTS**

3,996,345 A 12/1976 Ullman et al. .... 424/12  
4,637,988 A 1/1987 Hinshaw et al. .... 436/546

(Continued)

**FOREIGN PATENT DOCUMENTS**

EP 0 990 903 4/2000 ..... G01N/33/58  
WO WO 98/04740 2/1998 ..... C12Q/1/68  
WO WO 99/19515 4/1999 ..... C12Q/1/68

**OTHER PUBLICATIONS**

Alivisatos, A. P., "Semiconductor Clusters, Nanocrystals, and Quantum Dots," *Science* 271 (Feb. 16, 1996):933-937.

(Continued)

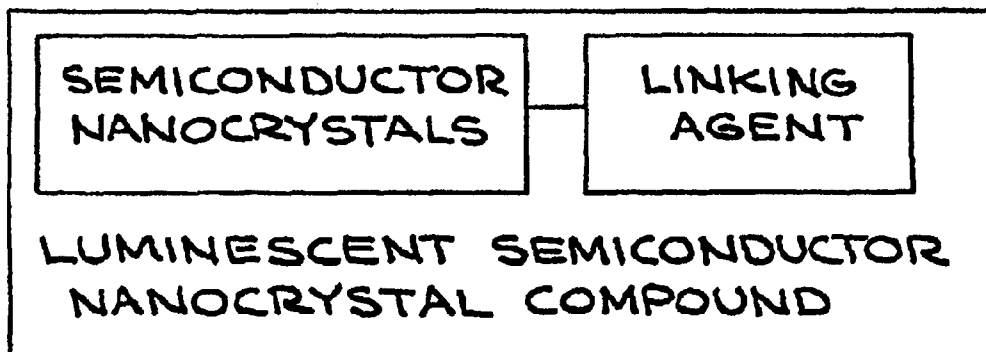
*Primary Examiner*—Christopher L. Chin

(74) *Attorney, Agent, or Firm*—Karl Bozicevic; Bozicevic, Field & Francis LLP

(57) **ABSTRACT**

A semiconductor nanocrystal compound is described capable of linking to an affinity molecule. The compound comprises (1) a semiconductor nanocrystal capable of emitting electromagnetic radiation and/or absorbing energy, and/or scattering or diffracting electromagnetic radiation—when excited by an electromagnetic radiation source or a particle beam; and (2) at least one linking agent, having a first portion linked to the semiconductor nanocrystal and a second portion capable of linking to an affinity molecule. The compound is linked to an affinity molecule to form a semiconductor nanocrystal probe capable of bonding with a detectable substance. Subsequent exposure to excitation energy will excite the semiconductor nanocrystal in the probe causing the emission of electromagnetic radiation. Further described are processes for respectively: making the luminescent semiconductor nanocrystal compound; making the semiconductor nanocrystal probe; and using the probe to determine the presence of a detectable substance in a material.

**23 Claims, 3 Drawing Sheets**



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## U.S. PATENT DOCUMENTS

4,777,128	A	10/1988	Lippa .....	435/5
5,262,357	A	11/1993	Alivisatos et al. ....	437/233
5,319,209	A	6/1994	Miyakawa et al. ....	250/459.1
5,460,831	A	* 10/1995	Kossovsky et al. ....	424/493
5,505,928	A	4/1996	Alivisatos et al. ....	423/299
5,537,000	A	7/1996	Alivisatos et al. ....	313/506
5,585,640	A	12/1996	Huston et al. ....	250/483.1
5,674,698	A	10/1997	Zarling et al. ....	435/7.92
5,736,330	A	4/1998	Fulton .....	435/6
5,751,018	A	5/1998	Alivisatos et al. ....	257/64
5,990,479	A	* 11/1999	Weiss et al. ....	436/172
6,023,540	A	2/2000	Walt et al.	
6,322,901	B1	* 11/2001	Bawendi et al. ....	428/548
6,423,551	B1	* 7/2002	Weiss et al. ....	436/518

## OTHER PUBLICATIONS

Alivisatos, A. P., "Perspectives on the Physical Chemistry of Semiconductor Nanocrystals," *J. Phys. Chem.* 100 (1996):13226–13239.

Alivisatos, A. Paul, et al., "Organization of 'Nanocrystal Molecules' Using DNA," *Nature* 382 (Aug. 15, 1996):609–611.

Beverloo, H.B., et al., "Preparation and Microscopic Visualization of Multicolor Luminescent Immunophosphors," Chapter 4 of Beverloo, H.B., "Inorganic Crystals as Luminescent Labels: Their Applications in Immunocytochemistry and Time-Resolved Microscopy," Ph.D. dissertation, University of Leiden (The Netherlands), May 13, 1992, pp. 553–573.

Bruchez, Marcel P., Jr., "Luminescent Semiconductor Nanocrystals: Intermittent Behavior and Use as Fluorescent Biological Probes," Ph.D. dissertation, University of California, Dec. 17, 1998.

Bruchez, Marcel, Jr., et al., "Semiconductor Nanocrystals as Fluorescent Probes for Biology," *Cytometry Suppl.* 9 (1998):26.

Chan, Warren C.W., et al., "Quantum Dot Bioconjugates for Ultrasensitive Nonisotopic Detection," *Science* 281 (Sep. 25, 1998):2016–2018.

Coffer, Jeffrey L., et al., "Characterization of Quantum-Confined CdS Nanocrystallites Stabilized by Deoxyribonucleic Acid (DNA)," *Nanotechnol.* 3 (1992):69–76.

Cook, Neil D., "Scintillation Proximity Assay: A Versatile High-Throughput Screening Technology," *Drug Discovery Today* 1 (Jul., 1996):287–294.

Correa-Duarte, Miguel A., et al., "Stabilization of CdS Semiconductor Nanoparticles Against Photodegradation by a Silica Coating Procedure," *Chem. Phys. Lett.* 286 (Apr. 17, 1998):497–501.

Jacoby, Mitch, "Quantum Dots Meet Biomolecules," *C&E News* 76 (Sep. 28, 1998):Copied from the Internet as pp. 1–3.

Kagan, C.R., et al., "Electronic Energy Transfer in CdSe Quantum Dot Solids," *Phys. Rev. Lett.* 76 (Feb. 26, 1996):1517–1520.

Leff, David N., "Color-Coding Quantum Dots Debut with Promising Careers in Clinical Diagnostics Field," *Bioworld Today*, Sep. 25, 1998, Copied from the Internet as pp. 1–2.

Liz-Marzán, Luis M., et al., "Synthesis of Nanosized Gold-Silica Core-Shell Particles," *Lanquair* 12 (1996):4329–4335.

Mahtab, Rahina, et al., "Preferential Adsorption of a 'Kinked' DNA to a Neutral Curved Surface: Comparisons to and Implications for Nonspecific DNA-Protein Interactions," *J. Am. Chem. Soc.* 118 (1996):7028–7032.

Mahtab, Rahina, et al., "Protein-Sized Quantum Dot Luminescence Can Distinguish Between 'Straight,' 'Bent,' and 'Kinked' Oligonucleotides," *J. Am. Chem. Soc.* 117 (1995):9099–9100.

Murphy, Catherine J., et al., "Quantum Dots as Inorganic DNA-Binding Proteins," *Mat. Res. Soc. Symp. Proc.* 452 (1997):597–600.

Peng, Xiaogang, et al., "Synthesis and Isolation of a Homodimer of Cadmium Selenide Nanocrystals," *Angewandte Chemie-International Edition in English*, 36 (1997):145–147.

Service, Robert F., "Semiconductor Beacons Light Up Cell Structures," *Science* 281 (Sep. 25, 1998):1930–1931.

Shröck, E., et al., "Multicolor Spectral Karyotyping of Human Chromosomes," *Science* 273 (Jul. 26, 1996):494–497.

Zhang, Yu-zhong, et al., "Novel Flow Cytometry Compensation Standards: Internally Stained Fluorescent Microspheres with Matched Emission Spectra and Long-Term Stability," *Cytometry* 33 (1998):244–248.

Lacoste, T.D., et al., "Super Resolution Molecular Ruler Using Single Quantum Dots," *Biophysical Journal*, vol. 78, Jan., 2000, p. 402A, XP-000933548 Abstract.

Bruchez, Marcel, Jr., et al., "Semiconductor Nanocrystals as Fluorescent Biological Labels," *Science*, vol. 281, Sep. 25, 1998, pp. 2013–2016.

Dabbousi, B.O., et al., "(CdSe)ZnS Core-Shell Quantum Dots: Synthesis and Characterization of a Size Series of Highly Luminescent Nanocrystal-lites," *Journal of Physical Chemistry B*, vol. 101, 1997, pp. 9463–9475.

Peng, Xiaogang, et al., "Epitaxial Growth of Highly Luminescent CdSe/CdS Core/Shell Nanocrystals with Photostability and Electronic Accessibility," *Journal of the American Chemical Society*, vol. 119, No. 30, pp. 7019–7029.

\* cited by examiner

FIG. 1

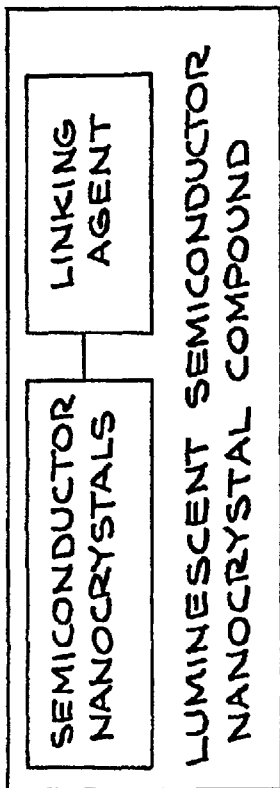


FIG. 2

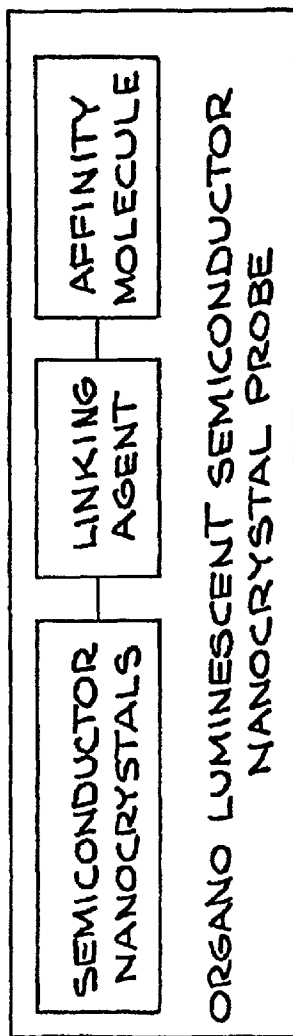
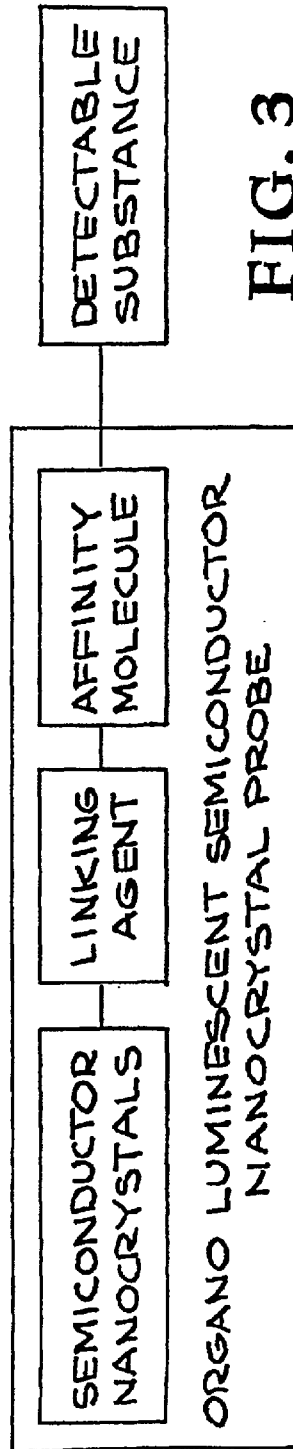


FIG. 3



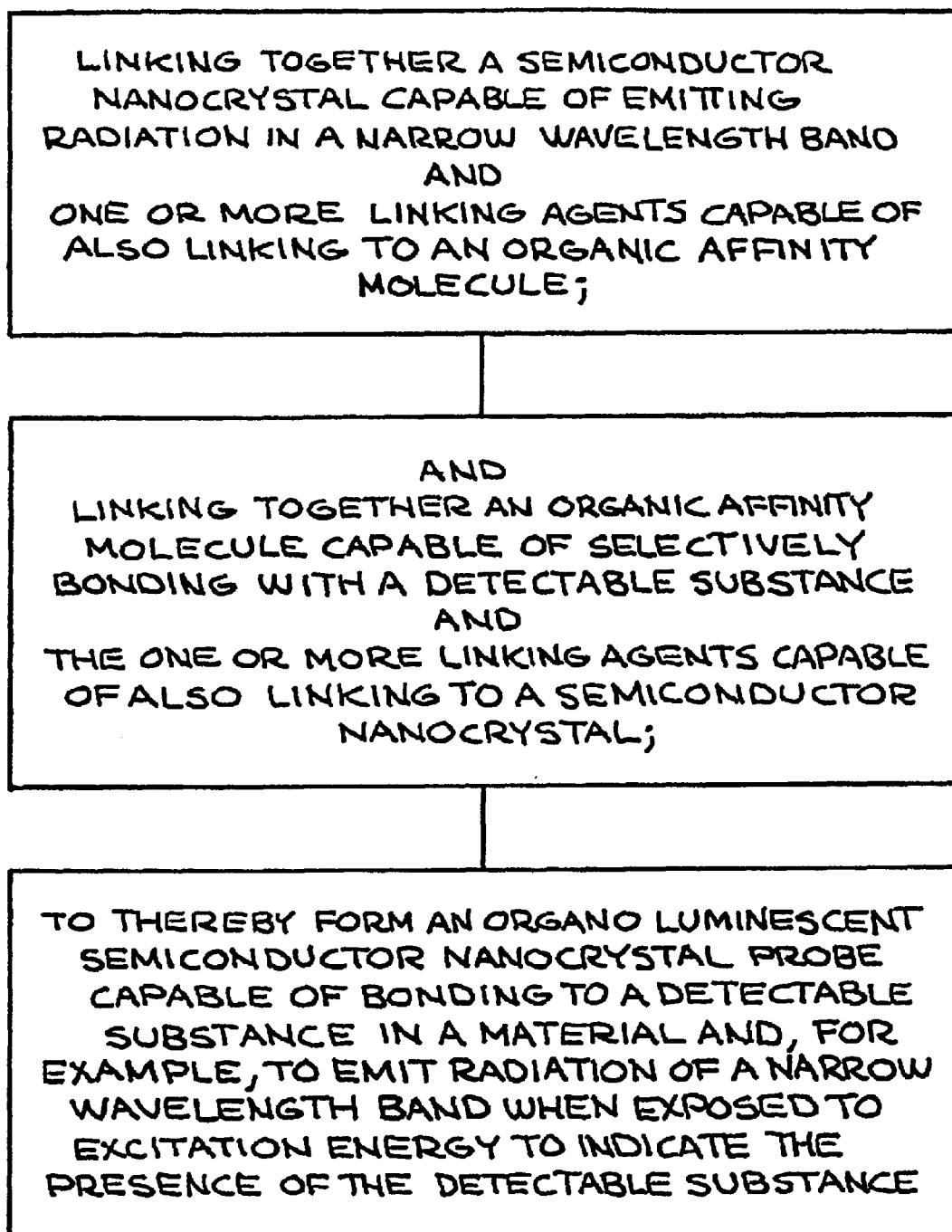


FIG. 4

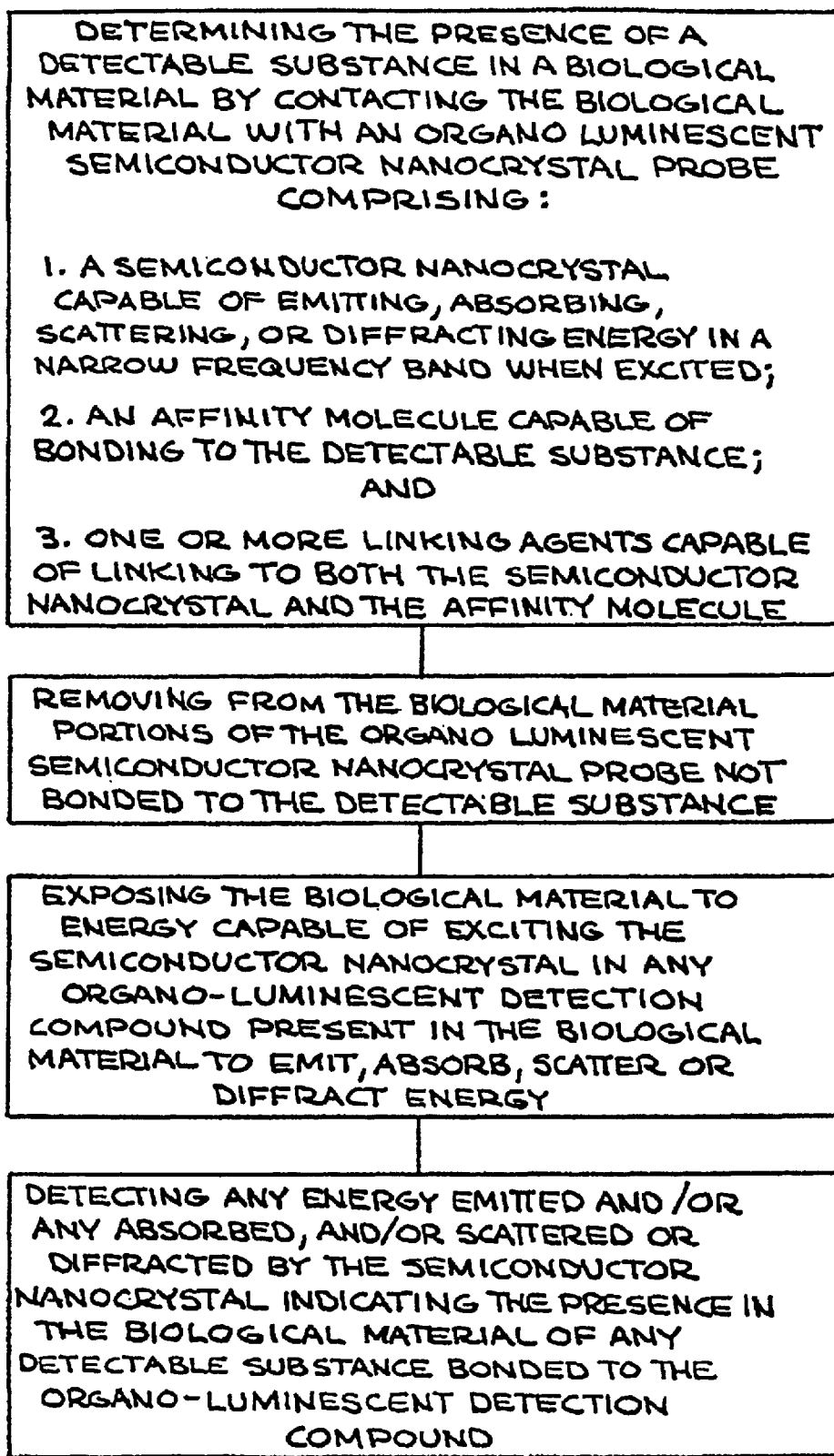


FIG. 5

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**ORGANO LUMINESCENT  
SEMICONDUCTOR NANOCRYSTAL PROBES  
FOR BIOLOGICAL APPLICATIONS AND  
PROCESS FOR MAKING AND USING SUCH  
PROBES**

CROSS-REFERENCE TO RELATED  
APPLICATION

This application is a continuation of U.S. patent application Ser. No. 09/349,833 filed Jul. 8, 1999 now U.S. Pat. No. 6,423,551 which application is a continuation of U.S. patent application Ser. No. 08/978,450 filed Nov. 25, 1997, and now issued as U.S. Pat. No. 5,990,479 on Nov. 23, 1999.

The invention described herein arose in the course of, or under, Contract No. DE-AC03-SF00098 between the United States Department of Energy and the University of California for the operation of the Ernest Orlando Lawrence Berkeley National Laboratory. The Government may have rights to the invention.

BACKGROUND OF THE INVENTION

1. Field of the Invention

This invention relates to organo luminescent semiconductor nanocrystal probes for biological applications wherein the probes includes a plurality of semiconductor nanocrystals capable of luminescence and/or absorption and/or scattering or diffraction when excited by a radiation or particle beam.

2. Description of the Related Art

Fluorescent labeling of biological systems is a well known analytical tool used in modern bio-technology as well as analytical chemistry. Applications for such fluorescent labeling include technologies such as medical (and non-medical) fluorescence microscopy, histology, flow cytometry, fluorescence in-situ hybridization (medical assays and research), DNA sequencing, immuno-assays, binding assays, separation, etc.

Conventionally, such fluorescent labeling involves the use of an organic dye molecule bonded to a moiety which, in turn, selectively bonds to a particular biological system, the presence of which is then identified by excitation of the dye molecule to cause it to fluoresce. There are a number of problems with such an analytical system. In the first place, the emission of light of visible wavelengths from an excited dye molecule usually is characterized by the presence of a broad emission spectrum as well as a broad tail of emissions on the red side of the spectrum, i.e., the entire emission spectrum is rather broad. As a result, there is a severe limitation on the number of different color organic dye molecules which may be utilized simultaneously or sequentially in an analysis since it is difficult to either simultaneously or even non-simultaneously detect or discriminate between the presence of a number of different detectable substances due to the broad spectrum emissions and emission tails of the labelling molecules. Another problem is that most dye molecules have a relatively narrow absorption spectrum, thus requiring either multiple excitation beams used either in tandem or sequentially for multiple wavelength probes, or else a broad spectrum excitation source which is sequentially used with different filters for sequential excitation of a series of probes respectively excited at different wavelengths.

Another problem frequently encountered with existing dye molecule labels is that of photostability. Available fluorescent molecules bleach, or irreversibly cease to emit light, under repeated excitation ( $10^4$ - $10^8$ ) cycles of absorption/emission. These problems are often surmounted by minimizing the amount of time that the sample is exposed

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to light, and by removing oxygen and/or other radical species from the sample.

In addition, the probe tools used for the study of these systems by electron microscopy techniques are completely different from the probes used for study by fluorescence. Thus, it is not possible to label a material with a single type of probe for both electron microscopy and for fluorescence.

It would, therefore, be desirable to provide a stable probe material for biological applications having a wide absorption band and capable of exhibiting either a detectable change in absorption or of emitting radiation in a narrow wavelength band, without the presence of the large red emission tails characteristic of dye molecules (thereby permitting the simultaneous use of a number of such probe materials, each emitting light of a different narrow wavelength band) and/or capable of scattering or diffracting radiation. It would also be equally desirable to provide a single, stable probe material which can be used to image the same sample by both light and electron microscopy.

SUMMARY OF THE INVENTION

The invention comprises a luminescent semiconductor nanocrystal compound capable of linking to an affinity molecule to form an organo luminescent semiconductor nanocrystal probe capable of luminescence and/or absorption and/or scattering or diffracting when excited by an electromagnetic radiation source (of broad or narrow bandwidth) or a particle beam, and capable of exhibiting a detectable change in absorption and/or of emitting radiation in a narrow wavelength band and/or scattering or diffracting when so excited. The luminescent semiconductor nanocrystal compound preferably comprises: (1) a semiconductor nanocrystal capable of luminescence and/or absorption and/or scattering or diffraction when excited by an electromagnetic radiation source (of broad or narrow bandwidth) or a particle beam, and capable of exhibiting a detectable change in absorption and/or of emitting radiation in a narrow wavelength band and/or scattering or diffracting when excited; and (2) a linking agent having a first portion linked to the semiconductor nanocrystal, and a second portion capable of linking to an affinity molecule.

The invention further comprises an organo luminescent semiconductor nanocrystal probe formed by linking the above described luminescent semiconductor nanocrystal compound to an affinity molecule capable of bonding to a detectable substance in a material. As a result the organo luminescent semiconductor nanocrystal probe, in one embodiment, is capable of absorbing or scattering or diffracting energy from either a particle beam or an electromagnetic radiation source (of broad or narrow bandwidth), and is capable of emitting electromagnetic radiation in a narrow wavelength band when so excited; while in another embodiment the amount of energy so absorbed, or scattered, or diffracted from either a particle beam or an electromagnetic radiation source (of broad or narrow bandwidth), is detectable, i.e., the change in absorption, scattering, or diffraction is detectable.

Therefore, treatment of a material with the organo luminescent semiconductor nanocrystal probe, and subsequent exposure of this treated material to excitation energy (from either a particle beam or an electromagnetic radiation source of broad or narrow bandwidth) to determine the presence of the detectable substance within the material, will excite the semiconductor nanocrystals in the organo luminescent semiconductor nanocrystal probe bonded to the detectable substance, resulting in the emission of electromagnetic radiation of a narrow wavelength band and/or a detectable change in the amount of energy being absorbed and/or scattered or diffracted, signifying the presence, in the



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material, of the detectable substance bonded to the organo luminescent semiconductor nanocrystal probe.

The invention also comprises a process for making the luminescent semiconductor nanocrystal compound and for making the organo luminescent semiconductor nanocrystal probe comprising the luminescent semiconductor nanocrystal compound linked to an affinity molecule capable of bonding to a detectable substance. The organo luminescent semiconductor nanocrystal probe of the invention is stable with respect to repeated excitation by light, or exposure to oxygen or other radicals. The invention further comprises a process for treating a material, such as a biological material, to determine the presence of a detectable substance in the material which comprises contacting the material with the organo luminescent semiconductor nanocrystal probe, removing from the material portions of the organo luminescent semiconductor nanocrystal probe not bonded to the detectable substance, and then exposing the material to activation energy from either an electromagnetic radiation source (of broad or narrow bandwidth) or a particle beam. The presence of the detectable substance in the material is then determined either by measuring the absorption of energy by the organo luminescent semiconductor nanocrystal probe and/or detecting the emission of radiation of a narrow wavelength band by the organo luminescent semiconductor nanocrystal probe and/or detecting the scattering or diffraction by the organo luminescent semiconductor nanocrystal probe, indicative (in either case) of the presence of the organo luminescent semiconductor nanocrystal probe bonded to the detectable substance in the material.

#### BRIEF DESCRIPTION OF THE DRAWINGS

FIG. 1 is a block diagram of the luminescent semiconductor nanocrystal compound of the invention.

FIG. 2 is a block diagram of the organo luminescent semiconductor nanocrystal probe of the invention.

FIG. 3 is a block diagram showing the affinity between a detectable substance and the organo luminescent semiconductor nanocrystal probe of the invention.

FIG. 4 is a flow sheet illustrating the process of forming the organo luminescent semiconductor nanocrystal probe of the invention.

FIG. 5 is a flow sheet illustrating a typical use of the organo luminescent semiconductor nanocrystal probe of the invention in detecting the presence of a detectable substance in a material such as a biological material.

#### DETAILED DESCRIPTION OF THE INVENTION

The invention comprises a luminescent semiconductor nanocrystal compound capable of linking to an organic molecule and capable of exhibiting a detectable change in absorption and/or of emitting electromagnetic radiation in a narrow wavelength band and/or scattering or diffracting when excited by either an electromagnetic radiation source (of broad or narrow bandwidth) or a particle beam. The luminescent semiconductor nanocrystal compound, in turn, comprises: (1) semiconductor nanocrystals capable of exhibiting a detectable change in absorption and/or of emitting electromagnetic radiation in a narrow wavelength band when excited by either an electromagnetic radiation source (of broad or narrow bandwidth) or a particle beam; and (2) one or more linking agents each having a first portion linked to the semiconductor nanocrystal and a second portion capable of linking to an organic affinity molecule.

The invention also comprises the above described luminescent semiconductor nanocrystal compound linked to the organic affinity molecule (through the linking agent) to form

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an organo luminescent semiconductor nanocrystal probe capable of bonding to a detectable substance and capable of exhibiting a detectable change in absorption and/or of emitting electromagnetic radiation in a narrow wavelength band and/or scattering or diffracting when excited by either an electromagnetic radiation source (of broad or narrow bandwidth) or a particle beam. Treatment of a material (typically a biological material) with the organo luminescent semiconductor nanocrystal probe, and subsequent exposure of this treated material to excitation energy, as described above, to determine the presence of the detectable substance within the material, will excite the semiconductor nanocrystal in the organo luminescent semiconductor nanocrystal probe bonded to the detectable substance, causing the detectable change in absorption and/or emission of electromagnetic radiation of a narrow wavelength band and/or scattering or diffraction signifying (in either instance) the presence in the material, of the detectable substance bonded to the organo luminescent semiconductor nanocrystal probe.

The invention also comprises a process for making the luminescent semiconductor nanocrystal compound, and a process for making the organo luminescent semiconductor nanocrystal probe comprising the luminescent semiconductor nanocrystal compound linked to an affinity molecule capable of bonding to a detectable substance.

The invention further comprises a process for treating a material, such as a biological material, to determine the presence of a detectable substance in the material which comprises: (1) contacting the material with the organo luminescent semiconductor nanocrystal probe, (2) removing from the material portions of the organo luminescent semiconductor nanocrystal probe not bonded to the detectable substance, (3) exposing the material to energy (such as the above-described electromagnetic energy source or particle beam) capable of exciting the semiconductor nanocrystal to cause a detectable change in absorption and/or emission of electromagnetic radiation of a narrow wavelength band and/or scattering or diffraction signifying (in either instance) the presence of the organo luminescent semiconductor nanocrystal probe bonded to the detectable substance in the material, and (4) detecting either the change in absorbed energy or the electromagnetic radiation emitted or the scattering or diffraction by the semiconductor nanocrystal in the organo luminescent semiconductor nanocrystal probe.

#### a. Definitions

By use of the terms "nanometer crystal" or "nanocrystal" herein is meant an organic or inorganic single crystal particle having an average cross-section no larger than about 20 nanometers (nm) or  $20 \times 10^{-9}$  meters (200 Angstroms), preferably no larger than about 10 nm (100 Angstroms) and a minimum average cross-section of about 1 nm, although in some instances a smaller average cross-section nanocrystal, i.e., down to about 0.5 nm (5 Angstroms), may be acceptable. Typically the nanocrystal will have an average cross-section ranging in size from about 1 nm (10 Angstroms) to about 10 nm (100 angstroms).

By use of the term "semiconductor nanocrystal" is meant a nanometer crystal or nanocrystal of Group II-VI and Group III-V semiconductor compounds capable of emitting electromagnetic radiation upon excitation, although the use of Group IV semiconductors such as germanium or silicon, or the use of organic semiconductors, may be feasible under certain conditions.

By use of the term "a narrow wavelength band", with regard to the electromagnetic radiation emission of the semiconductor nanocrystal, is meant a wavelength band of emissions not exceeding about 40 nm, and preferably not exceeding about 20 nm in width and symmetric about the

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center, in contrast to the emission bandwidth of about 100 nm for a typical dye molecule, with a red tail which may extend the band width out as much as another 100 nm. It should be noted that the bandwidths referred to are determined from measurement of the width of the emissions at half peak height (FWHM), and are appropriate in the range of 200 nm to 2000 nm.

By use of the term “a broad absorption band”, with regard to the electromagnetic radiation absorption of the semiconductor nanocrystal is meant a continuously increasing absorption from the onset, which occurs near to, but at slightly higher energy than the “narrow wavelength band” of the emission. This is in contrast to the “narrow absorption band” of dye molecules which occurs near the emission peak on the high energy side, but drops off rapidly away from that wavelength.

By use of the term “detectable substance” is meant an entity or group, the presence or absence of which in a material such as a biological material, is to be ascertained by use of the organo-luminescent semiconductor nanocrystal probe of the invention.

By use of the term “affinity molecule” is meant the portion of the organo luminescent semiconductor nanocrystal probe of the invention which will selectively bond to a detectable substance (if present) in the material (e.g., biological material) being analyzed.

By use of the term “linking agent” is meant a substance capable of linking with a semiconductor nanocrystal and also capable of linking to an affinity molecule.

The terms “link” and “linking” are meant to describe the adherence between the affinity molecule and the semiconductor nanocrystals, either directly or through a moiety identified herein as a linking agent. The adherence may comprise any sort of bond, including, but not limited to, covalent, ionic, hydrogen bonding, Van der Waals’ forces, or mechanical bonding, etc.

The terms “bond” and “bonding” are meant to describe the adherence between the affinity molecule and the detectable substance. The adherence may comprise any sort of bond, including, but not limited to, covalent, ionic, or hydrogen bonding, Van der Waals’ forces, or mechanical bonding, etc.

The term “luminescent semiconductor nanocrystal compound”, as used herein, is intended to define a semiconductor nanocrystal linked to one or more linking agents and capable of linking to an affinity molecule, while the term “organo-luminescent semiconductor nanocrystal probe” is intended to define a luminescent semiconductor nanocrystal compound linked to an affinity molecule.

The term “glass” as used herein is intended to include one or more oxides of silicon, boron, and/or phosphorus, or a mixture thereof, as well as the further optional inclusion of one or more metal silicates, metal borates or metal phosphates therein.

#### b. The Semiconductor Nanocrystals

The semiconductor nanocrystals useful in the practice of the invention include nanocrystals of Group II–VI semiconductors such as MgS, MgSe, MgTe, CaS, CaSe, CaTe, SrS, SrSe, SrTe, BaS, BaSe, BaTe, ZnS, ZnSe, ZnTe, CdS, CdSe, CdTe, HgS, HgSe, and HgTe; and nanocrystals of Group III–V semiconductors such as GaAs, InGaAs, InP, and InAs. As mentioned above, the use of Group IV semiconductors such as germanium or silicon, or the use of organic semiconductors, may also be feasible under certain conditions.

Formation of nanometer crystals of Group III–V semiconductors is described in copending and commonly

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assigned Alivisatos et al. U.S. Pat. No. 5,751,018; Alivisatos et al. U.S. Pat. No. 5,505,928; and Alivisatos et al. U.S. Pat. No. 5,262,357, which also describes the formation of Group II–VI semiconductor nanocrystals, and which is also assigned to the assignee of this invention. Also described therein is the, control of the size of the semiconductor nanocrystals during formation using crystal growth terminators. The teachings of Alivisatos et al. U.S. Pat. No. 5,751,018, and Alivisatos et al. U.S. Pat. No. 5,262,357 are each hereby specifically incorporated by reference.

In a preferred embodiment, the nanocrystals are used in a core/shell configuration wherein a first semiconductor nanocrystal forms a core ranging in diameter, for example, from about 20 Å to about 100 Å, with a shell of another semiconductor nanocrystal material grown over the core nanocrystal to a thickness of, for example, 1–10 monolayers in thickness. When, for example, a 1–10 monolayer thick shell of CdS is epitaxially grown over a core of CdSe, there is a dramatic increase in the room temperature photoluminescence quantum yield. Formation of such core/shell nanocrystals is described more fully in a publication by one of us with others entitled “Epitaxial Growth of Highly Luminescent CdSe/CdS Core/Shell Nanocrystals with Photostability and Electronic Accessibility”, by Peng, Schlamp, Kadavanich, and Alivisatos, published in the Journal of the American Chemical Society, Volume 119, No. 30. 1997, at pages 7019–7029, the subject matter of which is hereby specifically incorporated herein by reference.

The semiconductor nanocrystals used in the invention will have a capability of emitting light within a narrow wavelength band of about 40 nm or less, preferably about 20 nm or less, thus permitting the simultaneous use of a plurality of differently colored organo luminescent semiconductor nanocrystal probes with different semiconductor nanocrystals without overlap (or with a small amount of overlap) in wavelengths of emitted light (unlike the use of dye molecules with broad emission lines (e.g., ~100 nm) and broad tails of emission (e.g., another 100 nm) on the red side of the spectrum), thus allowing for the simultaneous detection of a plurality of detectable substances.

#### c. Affinity Molecule

The particular affinity molecule forming a part of the organo-luminescent semiconductor nanocrystal probe of the invention will be selected based on its affinity for the particular detectable substance whose presence or absence, for example, in a biological material, is to be ascertained. Basically, the affinity molecule may comprise any molecule capable of being linked to a luminescent semiconductor nanocrystal compound which is also capable of specific recognition of a particular detectable substance. In general, any affinity molecule useful in the prior art in combination with a dye molecule to provide specific recognition of a detectable substance will find utility in the formation of the organo-luminescent semiconductor nanocrystal probes of the invention. Such affinity molecules include, by way of example only, such classes of substances as monoclonal and polyclonal antibodies, nucleic acids (both monomeric and oligomeric), proteins, polysaccharides, and small molecules such as sugars, peptides, drugs, and ligands. Lists of such affinity molecules are available in the published literature such as, by way of example, the “Handbook of Fluorescent Probes and Research Chemicals”, (sixth edition) by R. P. Haugland, available from Molecular Probes, Inc.

#### d. The Linking Agent

The organo-luminescent semiconductor nanocrystal probe of the invention will usually find utility with respect to the detection of one or more detectable substances in

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organic materials, and in particular to the detection of one or more detectable substances in biological materials. This requires the presence, in the organo-luminescent semiconductor nanocrystal probe, of an affinity molecule or moiety, as described above, which will bond the organo-luminescent semiconductor nanocrystal probe to the detectable substance in the organic/biological material so that the presence of the detectable material may be subsequently ascertained. However, since the semiconductor nanocrystals are inorganic, they may not bond directly to the organic affinity molecule. In these case therefore, there must be some type of linking agent present in the organo-luminescent semiconductor nanocrystal probe which is capable of forming a link to the inorganic semiconductor nanocrystal as well as to the organic affinity molecule in the organo-luminescent semiconductor nanocrystal probe.

One form in which the semiconductor nanocrystal may be linked to an affinity molecule via a linking agent is by coating the semiconductor nanocrystal with a thin layer of glass, such as silica ( $\text{SiO}_x$  where  $x=1-2$ ), using a linking agent such as a substituted silane, e.g., 3-mercaptopropyl-trimethoxy silane to link the nanocrystal to the glass. The glass-coated semiconductor nanocrystal may then be further treated with a linking agent, e.g., an amine such as 3-aminopropyl-trimethoxysilane, which will function to link the glass-coated semiconductor nanocrystal to the affinity molecule. That is, the glass-coated semiconductor nanocrystal may then be linked to the affinity molecule. It is within the contemplation of this invention that the original luminescent semiconductor nanocrystal compound may also be chemically modified after it has been made in order to link effectively to the affinity molecule. A variety of references summarize the standard classes of chemistry which may be used to this end, in particular the "Handbook of Fluorescent Probes and Research Chemicals", (6th edition) by R. P.

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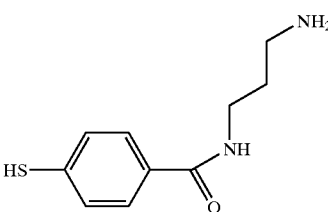
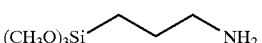
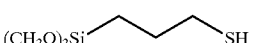
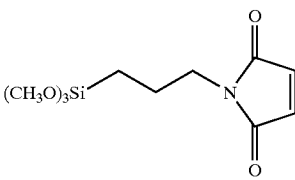
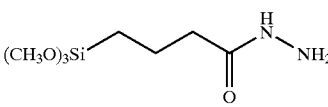
Haugland, available from Molecular Probes, Inc., and the book "Bioconjugate Techniques", by Greg Hermanson, available from Academic Press, New York.

When the semiconductor nanocrystal is coated with a thin layer of glass, the glass, by way of example, may comprise a silica glass ( $\text{SiO}_x$  where  $x=1-2$ ), having a thickness ranging from about 0.5 nm to about 10 nm, and preferably from about 0.5 nm to about 2 nm.

The semiconductor nanocrystal is coated with the coating of thin glass, such as silica, by first coating the nanocrystals with a surfactant such as tris-octyl-phosphine oxide, and then dissolving the surfactant-coated nanocrystals in a basic methanol solution of a linking agent, such as 3-mercaptopropyl-tri-methoxy silane, followed by partial hydrolysis which is followed by addition of a glass-affinity molecule linking agent such as amino-propyl trimethoxysilane which will link to the glass and serve to form a link with the affinity molecule.

When the linking agent does not involve the use of a glass coating on the semiconductor nanocrystal, it may comprise a number of different materials, depending upon the particular affinity molecule, which, in turn, depends upon the type of detectable material being analyzed for. It should also be noted that while an individual linking agent may be used to link to an individual semiconductor nanocrystal, it is also within the contemplation of the invention that more than one linking agent may bond to the same semiconductor nanocrystal and vice versa.

A few examples of the types of linking agents which may be used to link to both the semiconductor nanocrystal (or to a glass coating on the nanocrystal) and to the organic affinity molecule in the probe are illustrated in the table below, it being understood that this is not intended to be an exhaustive list:

<u>Linking Agent</u>	
Structure	Name
	N-(3-aminopropyl)3-mercapto-benzamide
	3-aminopropyl-trimethoxysilane
	3-mercaptopropyl-trimethoxysilane
	3-maleimidopropyl-trimethoxysilane
	3-hydrazidopropyl-trimethoxysilane

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It should be further noted that a plurality of polymerizable linking agents may be used together to form an encapsulating net or linkage around an individual nanocrystal (or group of nanocrystals). This is of particular interest where the particular linking agent is incapable of forming a strong bond with the nanocrystal. Examples of linking agents capable of bonding together in such a manner to surround the nanocrystal with a network of linking agents include, but are not limited to: diacetylenes, acrylates, acrylamides, vinyl, styryl, and the aforementioned silicon oxide, boron oxide, phosphorus oxide, silicates, borates and phosphates.

e. The Excitation of the Probe and Detection of Emission/Absorption

As previously mentioned, the organo luminescent semiconductor nanocrystal probe of the invention is capable of being excited over a broad bandwidth, yet exhibits emission in a narrow wavelength band, in contrast to the dye molecules used in the prior art. Thus electromagnetic radiation of wavelength ranging from x-ray to ultraviolet to visible to infrared waves may be used to excite the luminescent semiconductor nanocrystals in the probe. In addition, the luminescent semiconductor nanocrystals are capable of excitation from bombardment with a particle beam such as an electron beam (e-beam). Furthermore, because of the broad bandwidth at which the luminescent semiconductor nanocrystals are excitable, one may use a common excitation source for the simultaneous excitation of several probes, i.e., several probes which give off radiation at different frequencies, thus permitting simultaneous excitation and detection of the presence of several probes indicating, for example, the presence of several detectable substances in the material being examined.

Thus, for example, a laser radiation source of a given frequency, e.g., blue light, may be used to excite a first organo luminescent semiconductor nanocrystal probe capable of emitting radiation of a second frequency, e.g., red light, indicating the presence, in the material being illuminated, of a first detectable substance to which the particular red light-emitting organo luminescent semiconductor nanocrystal probe has bonded. At the same time, the same blue light laser source may also be exciting a second organo luminescent semiconductor nanocrystal probe (in the same material) capable of emitting radiation of a third frequency, e.g., green light, indicating the presence, in the material being illuminated, of a second detectable substance to which the particular green light-emitting organo luminescent semiconductor nanocrystal probe has bonded. Thus, unlike the prior art, multiple excitation sources need not be used (because of the broad bandwidth in which the organo luminescent semiconductor nanocrystal probe of the invention is capable of being excited), and the narrow band of emission of the specific semiconductor nanocrystals in each probe makes possible the elimination of sequencing and/or elaborate filtering to detect the emitted radiation.

With respect to the absorption of energy by the probe of the invention, when the excitation source is an electron beam, or an X-ray source, the presence of the organo luminescent semiconductor nanocrystal probe bonded to the detectable substance of interest in the material being analyzed can be ascertained using a commercially available energy absorption or scattering or diffraction detection system wherein changes in absorption or scattering cross section or in diffraction of the material being analyzed can be detected, signifying the presence of the probe in the material, which, in turn, indicates the presence of the detectable substance to which the probe is bonded in the material being analyzed. In addition, it may be possible to use electron or X-ray sources to detect the presence of the organo luminescent semiconductor nanocrystal probe

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bonded to the detectable substance by using a conventional detection system for the emission of visible light to observe the visible emission in the narrow wavelength of emission of the probe.

The following examples will serve to further illustrate the formation of the organo luminescent semiconductor nanocrystal probes of the invention, as well as their use in detecting the presence of a detectable substance in a material such as a biological material.

EXAMPLE 1

To illustrate the formation of the luminescent semiconductor nanocrystal compound (comprising the semiconductor nanocrystals linked to a linking agent) 20 ml. of a 5 mM solution of (4-mercapto)benzoic acid was prepared with a pH of 10 using  $(\text{CH}_3)_4\text{NOH} \cdot 0.5\text{H}_2\text{O}$ . 20 mg of tris-octylphosphine oxide coated CdSe/CdS core/shell nanocrystals were added to the solution and stirred until completely dissolved. The resultant nanocrystal/linking agent solution was heated for 5 hours at 50–60° C. and then concentrated to a few ml by evaporation. Then an equal volume of acetone was added and the nanocrystals precipitate out of solution homogeneously. The precipitate was then washed with acetone, dried, and then can be stored.

The luminescent semiconductor nanocrystal compound prepared above can be linked with an appropriate affinity molecule to form the organo luminescent semiconductor nanocrystal probe of the invention to treat a biological material to determine the presence or absence of a detectable substance. That is, the luminescent semiconductor nanocrystal compound prepared above can be linked, for example, with avidin or streptavidin (as the affinity molecule) to form an organo luminescent semiconductor nanocrystal probe to treat a biological material to ascertain the presence of biotin; or the luminescent semiconductor nanocrystal compound prepared above can be linked with anti-digoxigenin to form an organo luminescent semiconductor nanocrystal probe to treat a biological material to ascertain the presence of digoxigenin.

EXAMPLE 2

To illustrate the formation of luminescent semiconductor nanocrystal compound (comprising glass-coated semiconductor nanocrystals linked to a linking agent), 50  $\mu\text{l}$  of 3-mercaptopropyl-trimethoxy silane was added to 40 ml of an anhydrous solution of 25 vol. % dimethylsulfoxide in methanol, and the pH was adjusted to 10–11 using  $(\text{CH}_3)_4\text{NOH} \cdot 0.5\text{H}_2\text{O}$ . 10 mg of tris-octylphosphine oxide coated CdSe/CdS core-shell particles, prepared by the technique described in the aforementioned Peng, Schlamp, Kadavanich, and Alivisatos article, were then dissolved in this solution, and stirred for several hours. The solution was diluted with 40 ml of methanol adjusted to a pH of 10 with  $(\text{CH}_3)_4\text{NOH} \cdot 0.5\text{H}_2\text{O}$ , and heated for 1 hour at 69° C. The solution was stirred for an hour, and 40 ml of a 90 vol. % methanol/9.89 vol. %  $\text{H}_2\text{O}$ /0.1 vol. % trimethoxysilylpropyl urea/0.01 vol. % aminopropyl-trimethoxy silane solution which had been stirring for at least an hour, was added, and stirred for 2 hours. Subsequently the reaction was heated to 69° C. for 15 minutes, and then cooled. 10 ml of a 10 vol. % chlorotrimethyl silane solution in methanol which had been adjusted to a pH of 10 using  $(\text{CH}_3)_4\text{NOH} \cdot 0.5\text{H}_2\text{O}$  was added, stirred for 2 hours, then heated to 60° C. and then partially concentrated under vacuum. Once the methanol had all evaporated, the solution was precipitated with acetone as an oil product comprising the luminescent semiconductor nanocrystal compound. The luminescent semiconductor nanocrystal compound may then be redissolved in water, and in a variety of buffer solutions to prepare it for

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linking it to an affinity molecule to form the organo luminescent semiconductor nanocrystal probe of the invention to treat a biological material to determine the presence or absence of a detectable substance.

Thus, the invention provides an organo luminescent semiconductor nanocrystal probe containing a semiconductor nanocrystal capable, upon excitation by either electromagnetic radiation (of either narrow or broad bandwidth) or particle beam, of emitting electromagnetic radiation in a narrow wavelength band and/or absorbing energy and/or scattering or diffracting said excitation, thus permitting the simultaneous usage of a number of such probes emitting different wavelengths of electromagnetic radiation to thereby permit simultaneous detection of the presence of a number of detectable substances in a given material. The probe material is stable in the presence of light or oxygen, capable of being excited by energy over a wide spectrum, and has a narrow band of emission, resulting in an improved material and process for the simultaneous and/or sequential detection of a number of detectable substances in a material such as a biological material.

Having thus described the invention what is claimed is:

1. A probe, comprising:
  - (a) a semiconductor nanocrystal which emits light when excited;
  - (b) a linking agent, linked to the semiconductor nanocrystal; and
  - (c) an affinity molecule linked to the linking agent.
2. The probe of claim 1, wherein the affinity molecule is a biological material.
3. The probe of claim 1, wherein the affinity molecule is an antibody.
4. The probe of claim 3, wherein the antibody is a monoclonal antibody.
5. The probe of claim 3, wherein the antibody is a polyclonal antibody.
6. The probe of claim 1, wherein the affinity molecule is a nucleic acid.
7. The probe of claim 6, wherein the nucleic acid is monomeric.
8. The probe of claim 6, wherein the nucleic acid is oligomeric.
9. The probe of claim 1, wherein the affinity molecule is a protein.
10. The probe of claim 1, wherein the affinity molecule is a polysaccharide.

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11. The probe of claim 1, wherein the affinity molecule is a sugar.

12. The probe of claim 1, wherein the affinity molecule is a peptide.

13. The probe of claim 1, wherein the affinity molecule is a drug.

14. A coated structure, comprising:

a semiconductor nanocrystal core which emits light when excited; and

a coating comprised of silica glass positioned at least partially around the core.

15. The coated structure of claim 14, wherein the glass comprises silica glass represented by the formula  $\text{SiO}_x$ , wherein x is selected from the group consisting of 1 and 2.

16. The coated structure of claim 14, wherein the coating has a thickness in a range of from about 0.5 nm to about 10 nm.

17. The coated structure of claim 14, wherein the coating has a thickness in a range of from about 0.5 nm to about 2 nm.

18. A composition, comprising:

a semiconductor nanocrystal which emits light when excited;

a polymer; and

an affinity molecule.

19. The composition as claimed in claim 18, wherein the polymer encapsulates the semiconductor nanocrystal.

20. The composition as claimed in claim 18, further comprising a first additional semiconductor nanocrystal which emits light when excited.

21. The composition as claimed in claim 18, further comprising:

a plurality of additional semiconductor nanocrystals which emit light when excited.

22. A composition, comprising:

a plurality of semiconductor nanocrystals which emit light when excited;

a polymerizable linking agent encapsulating the nanocrystals; and

an affinity molecule.

23. The composition as claimed in claim 22 wherein the linking agent is comprised of a polymer and chosen from diacetylenes, acrylates, acrylamides, vinyl, and styryl.

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